



Sustainability & Materials

Prof. Tiffany Abitbol
2025



ALL THE METALS WE MINED IN 2021

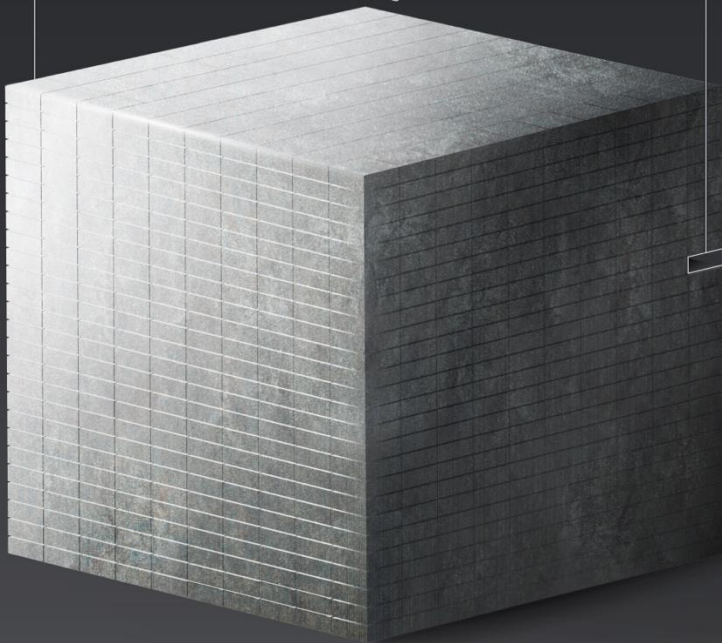
The world produced roughly **2.8 billion tonnes** of metals in 2021. Here are all the metals we mined, visualized on the same scale.

IRON ORE

2,600,000,000 tonnes*

= 1,000,000 tonnes

Iron Ore*
2.6B



LARGEST END-USE



Steelmaking



Construction



Chemicals



Alloying Agents



Energy/Batteries



Magnets



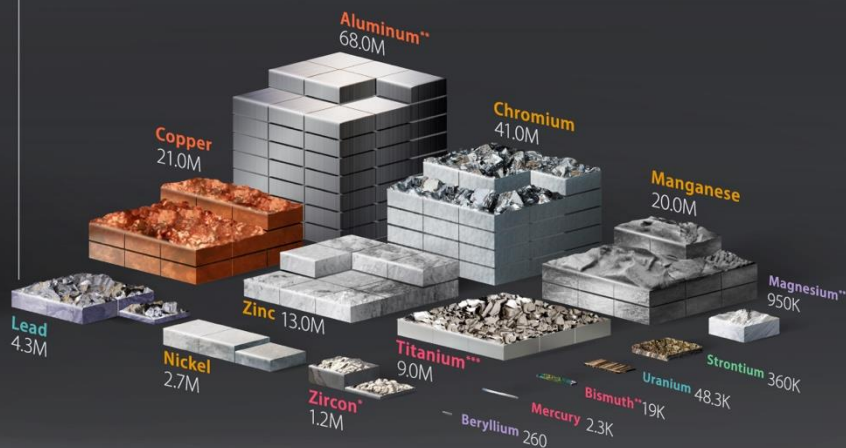
Electronics



Other

INDUSTRIAL METALS

181,579,892 tonnes



TECHNOLOGY AND PRECIOUS METALS

1,474,889 tonnes



Processing – Metals

After ore extraction, metal is refined and prepared for further processing.

Several possible techniques are available to produce metal parts, including:

- **Forging**, metal is heated to the point it can be shaped with the help of a hammer
- **Casting**, molten metal is poured into a mold and let cool down to solidify
- **Drawing**, a tensile force is applied to pull metal through a die (e.g. pipes)
- **Welding**, strong heat is used to join two or more individual pieces together
- **Machining**, subtractive shaping process used to remove unwanted material and refine the final form



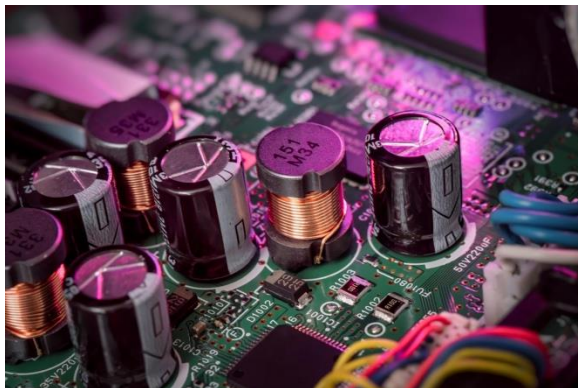
Properties & Uses – Metals

Main properties

- Versatility
- Strength
- Toughness
- Machinability
- Recyclability

Main uses

- Construction
- Food packaging
- Electronics
- Automotive
- Luxury goods

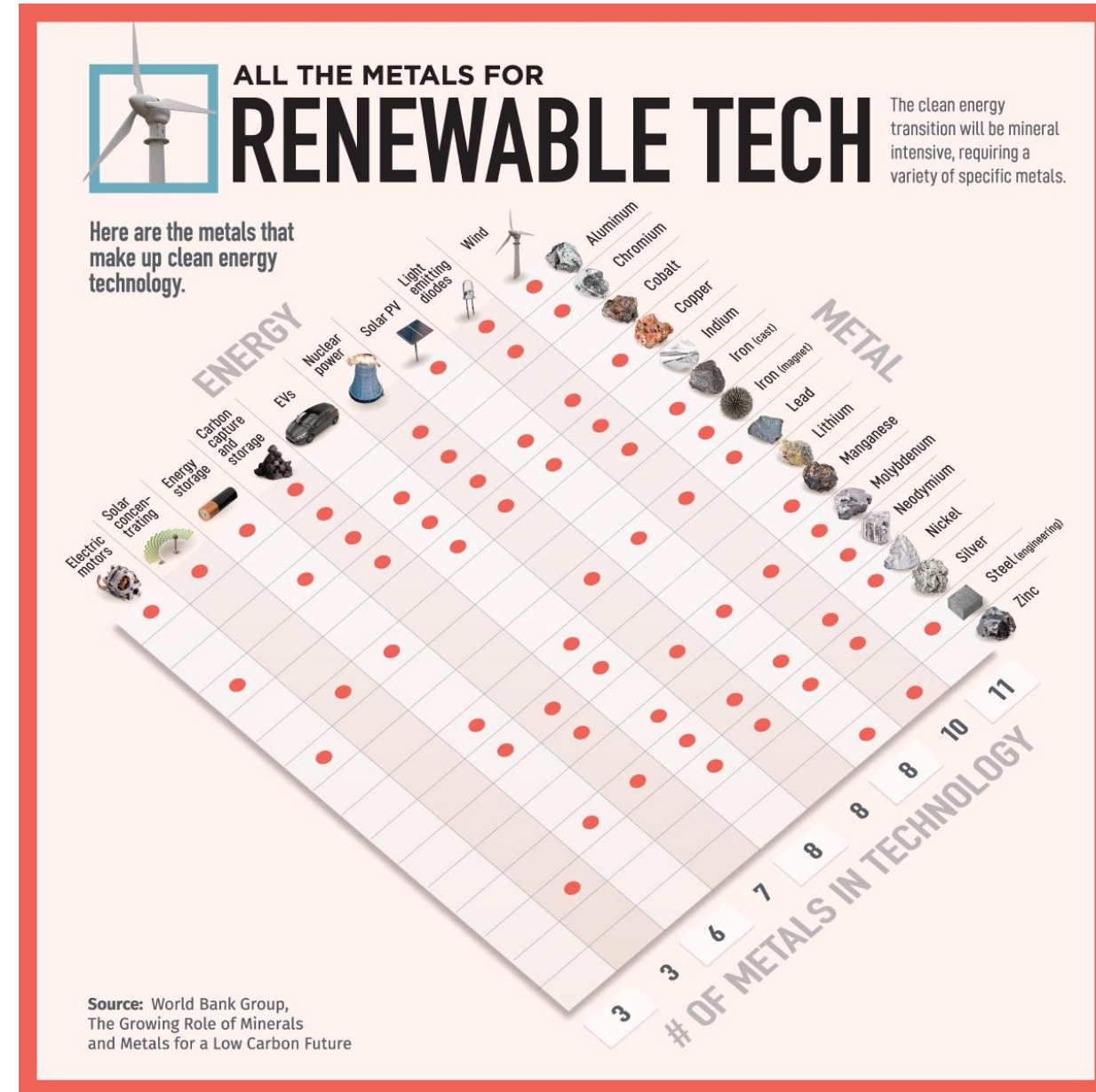


Metals and Clean Energy

Clean energy technology is heavily dependent on metals!

- **Electric cars** require 6x more metallic materials than conventional cars
- **Wind turbines** require 9x more metallic materials than gas-fired power plants

Inevitably, more mining will be required to meet the increasing demand of metallic materials for a renewable energy transition




9 - New tech, product design


A KEY CONTRIBUTOR TO THE CIRCULAR ECONOMY

One way to reduce waste generation is to ensure that resources and materials remain in the economy for as long as possible. As is communicated to the consumer through the Metal Recycles Forever logo, metal is infinitely recyclable without loss of quality, making metal packaging a key contributor to the circular economy. Metal packaging also has one of the highest recycling rates: 74.5% for aluminium cans and 82.5% for steel packaging.

12 RESPONSIBLE CONSUMPTION AND PRODUCTION



12 RESPONSIBLE CONSUMPTION AND PRODUCTION



VITAL TO REDUCING FOOD WASTE


The metal food can has a vital role to play in reducing global food waste and food losses. By using cans, we can reduce individual food waste by almost 30%. Their strength, barrier properties and versatility all help to keep food fresh for longer and minimise product loss during filling and transport.

12 - Metal food packaging to reduce waste, keep metals in circulation, recycle


INCREASING RESOURCE EFFICIENCY THROUGH INNOVATION

New technologies and policies have helped to increase resource-efficiency in the metal packaging industry, cutting CO2 emissions, saving energy and contributing to the circular economy. This includes constant innovation, re-thinking and re-shaping of products to reduce the amount of material required to make them. CO2 emissions from the production of aluminium drink cans, for example, decreased by an average of 31% over a ten year period.

9 INDUSTRY, INNOVATION AND INFRASTRUCTURE



13 CLIMATE ACTION



TRANSITIONING TO RENEWABLE ENERGY SOURCES

The metal packaging industry has put in place a range of initiatives aimed at significantly reducing greenhouse gas emissions, increasing reliance on renewable energy sources, such as wind and solar power, and cutting the carbon footprint across all operations. In addition, the industry continues to support efforts to increase already high recycling rates across Europe.

INCREASING WATER-USE EFFICIENCY

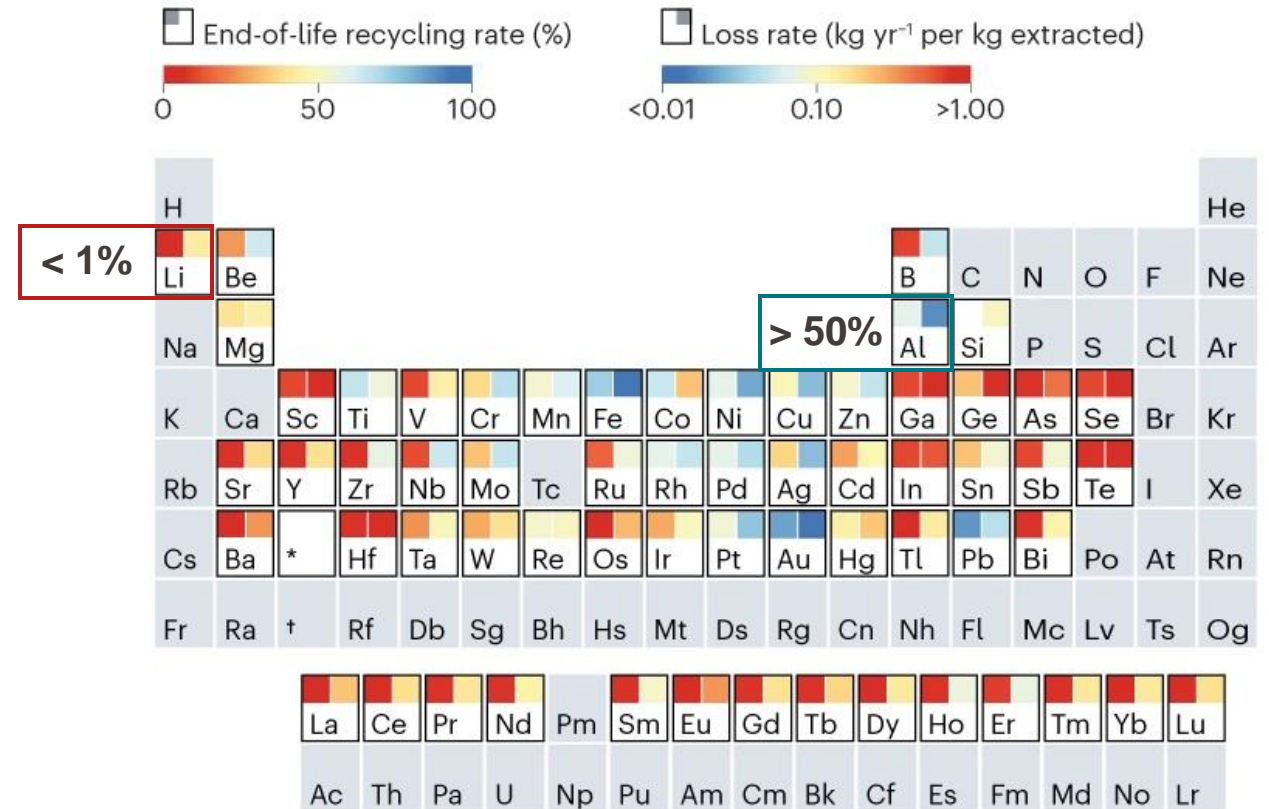
The metal packaging industry is committed to reducing water usage across its operations. It has achieved consistent improvements in water-use efficiency by developing and using new technologies, re-using water, consistently monitoring water data and sharing best practices.

6 CLEAN WATER AND SANITATION



13,6 – Renewable energy, efficient water use

- Reading the graph: color map for recycling rate (upper left) and loss rate (upper right)
- Loss can happen across life cycle, but usually lost at end of life to landfill or environment
- Strategic metals (e.g., lithium) have low rates of recycling (< 5% due to cost and complexity) and high rates of loss; compare to Al
- Poor end of life management can result in accumulation in the environment (ex. heavy metal pollution)



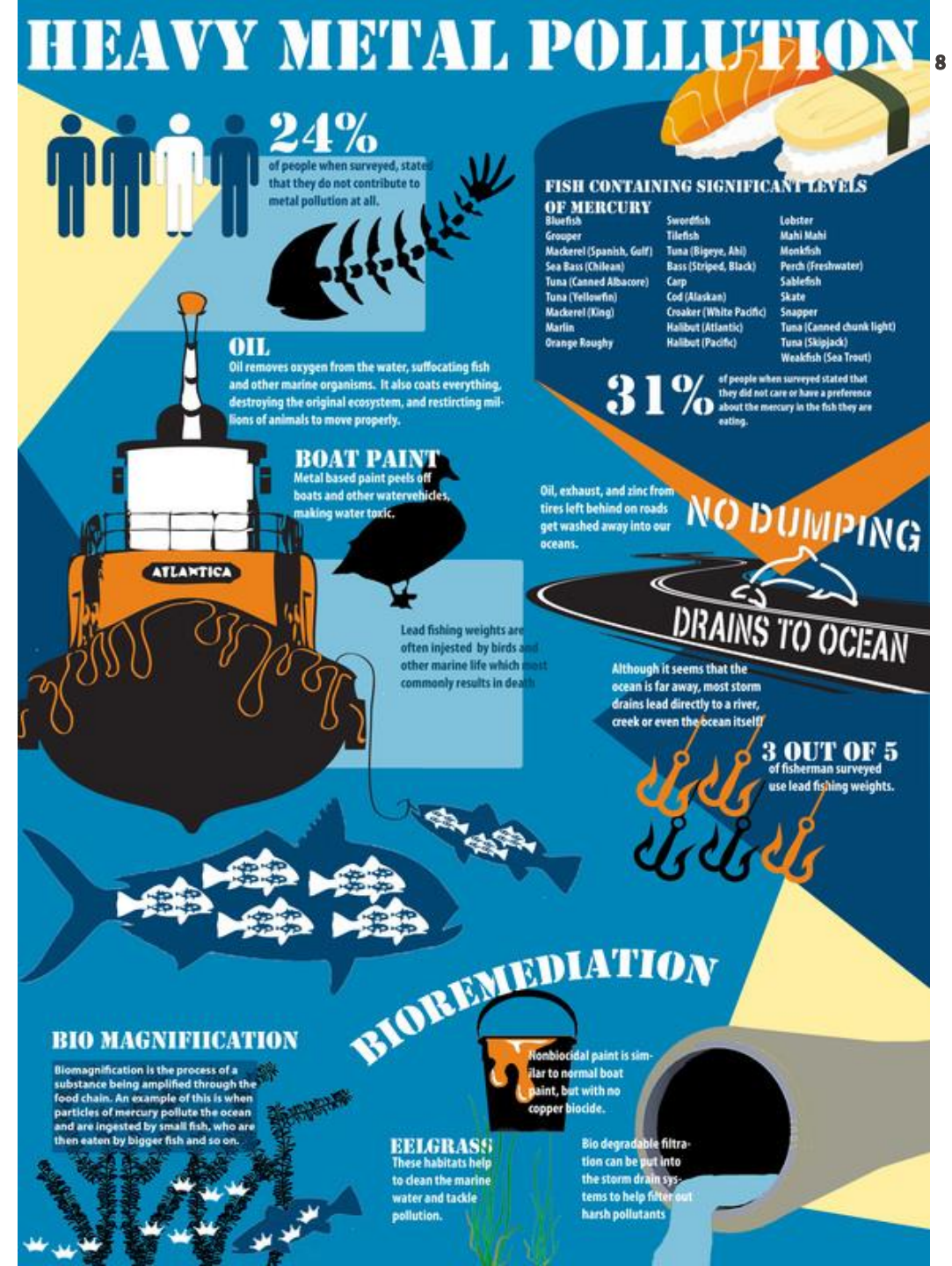
*Lanthanide series; *Actinide series.

©nature

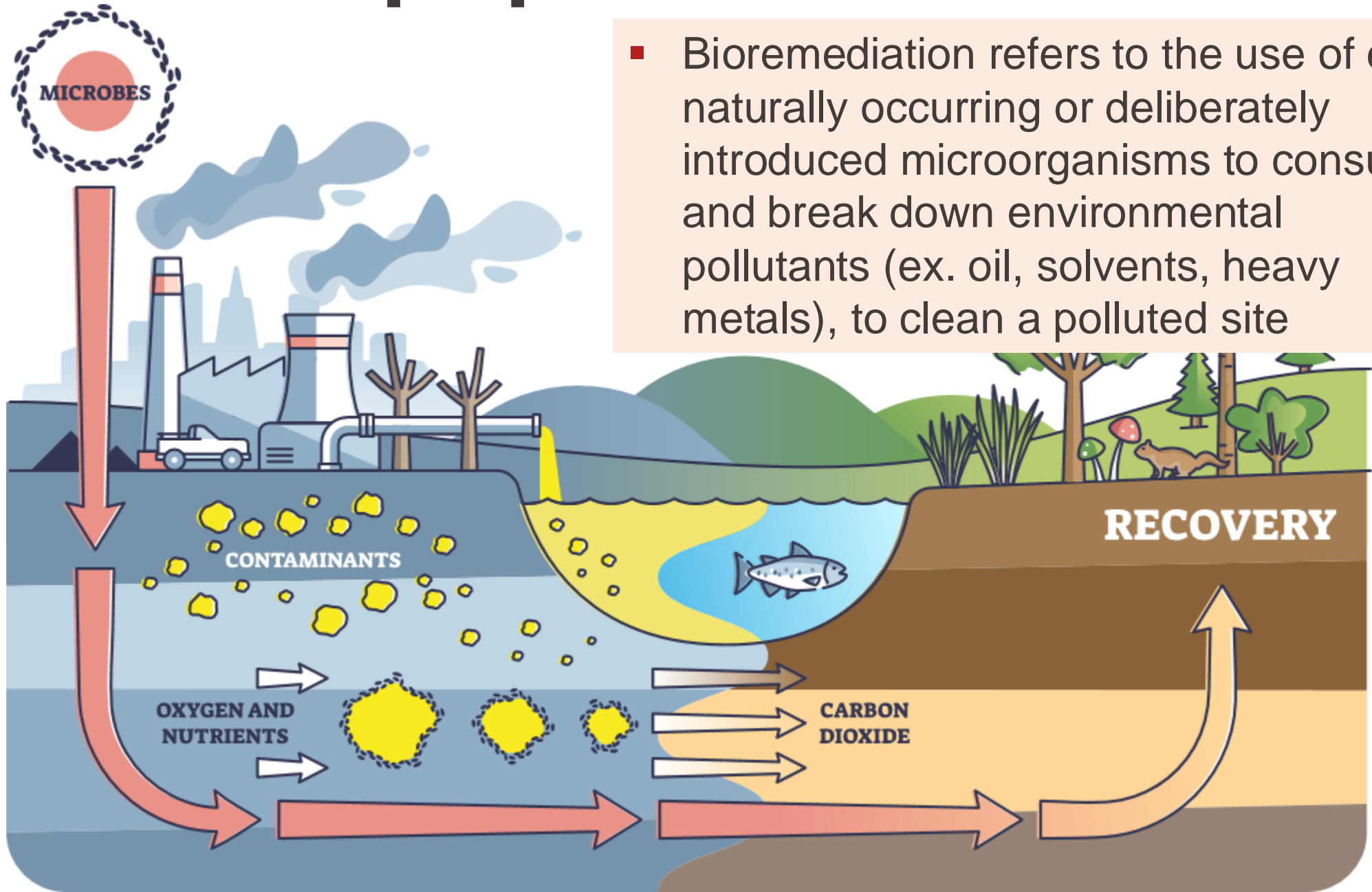
End of life – Metals

- Toxic metals, like mercury, can contaminate soil and water through natural events, like weathering or volcanic activities, or through manmade sources, such as improper waste disposal, industrial activities such as mining, agriculture, etc.,
- These metals can “bioaccumulate” up through the food chain

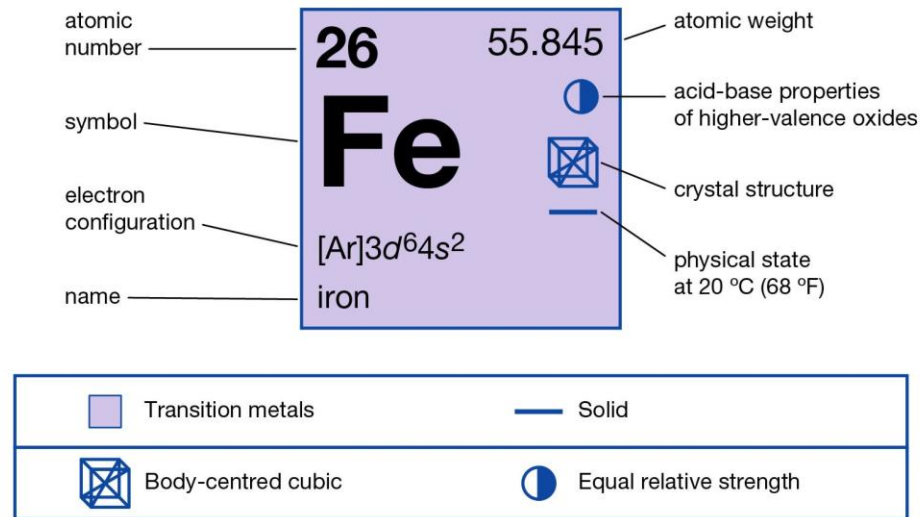
Image from San Diego Oceans Away Project



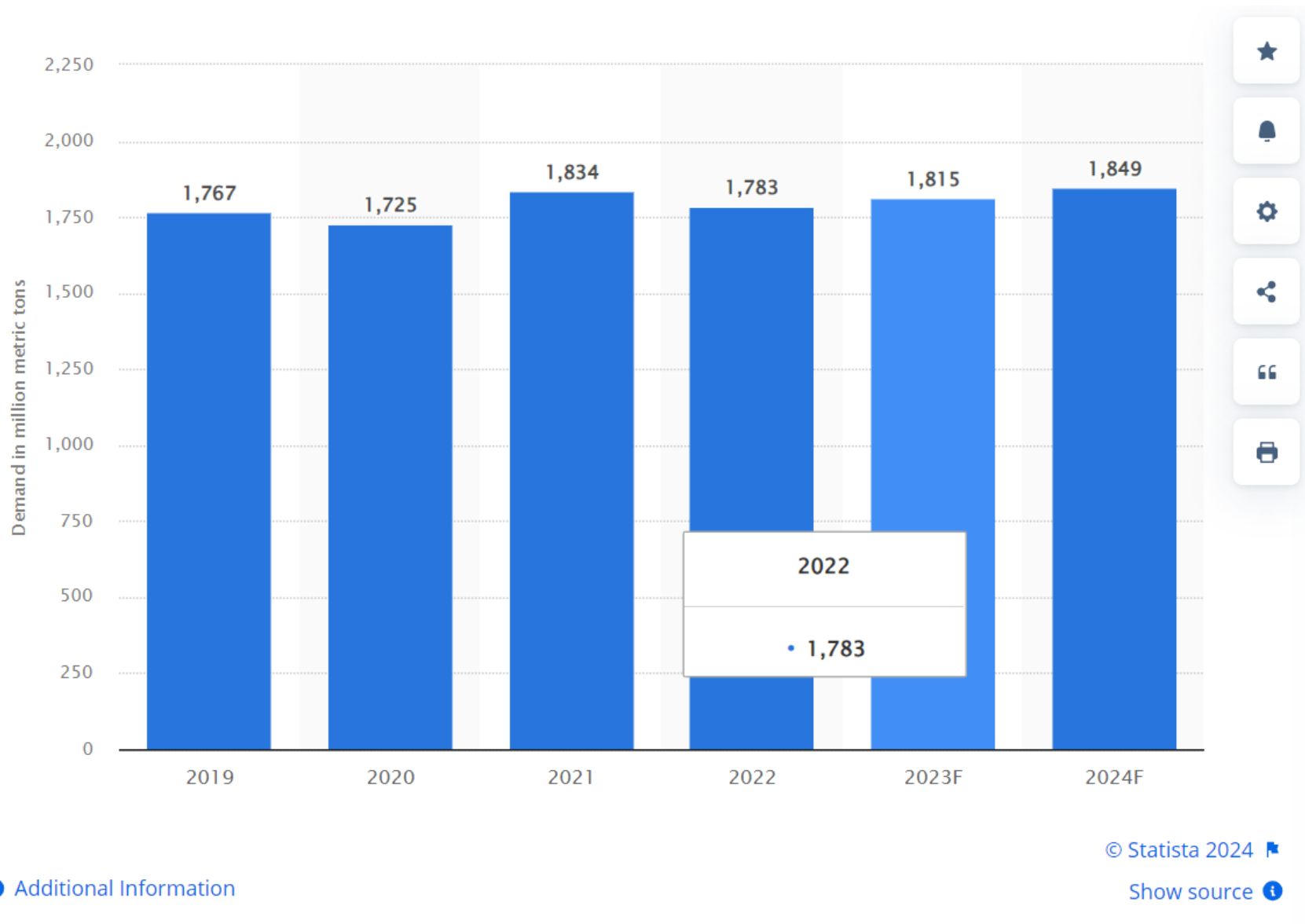
Sustainable perspectives – Bioremediation



- Bioremediation refers to the use of either naturally occurring or deliberately introduced microorganisms to consume and break down environmental pollutants (ex. oil, solvents, heavy metals), to clean a polluted site



- Iron is the 2nd most abundant metal in the crust (after aluminum)
- Iron ore makes up 93% of total metals mined (2021)
- First traces of iron objects were found in Egypt dated around 3400 BC
- Roughly 98% of iron ore goes into steelmaking
- Steel is an alloy of carbon and iron
- Critical to building infrastructure – key to urbanization and growth

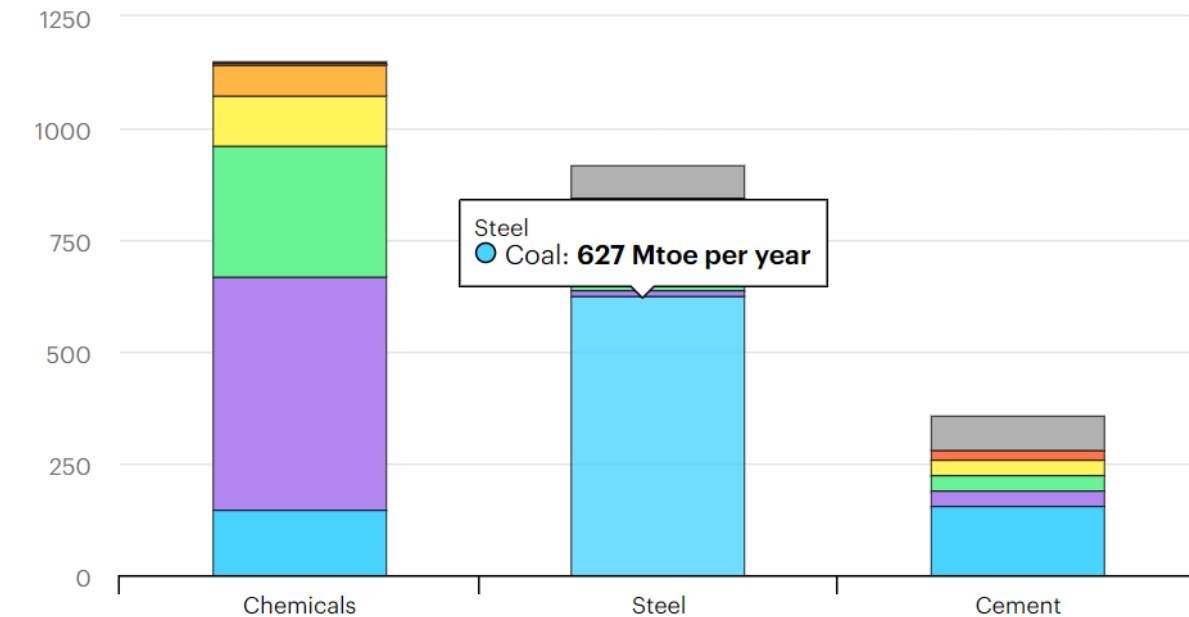


- Global steel demand is projected to increase by 1/3 by 2050 (from ca. 1.8 million metric tons now)

Final energy demand of selected heavy industry sectors by fuel, 2019

Open ↗

Mtoe per year

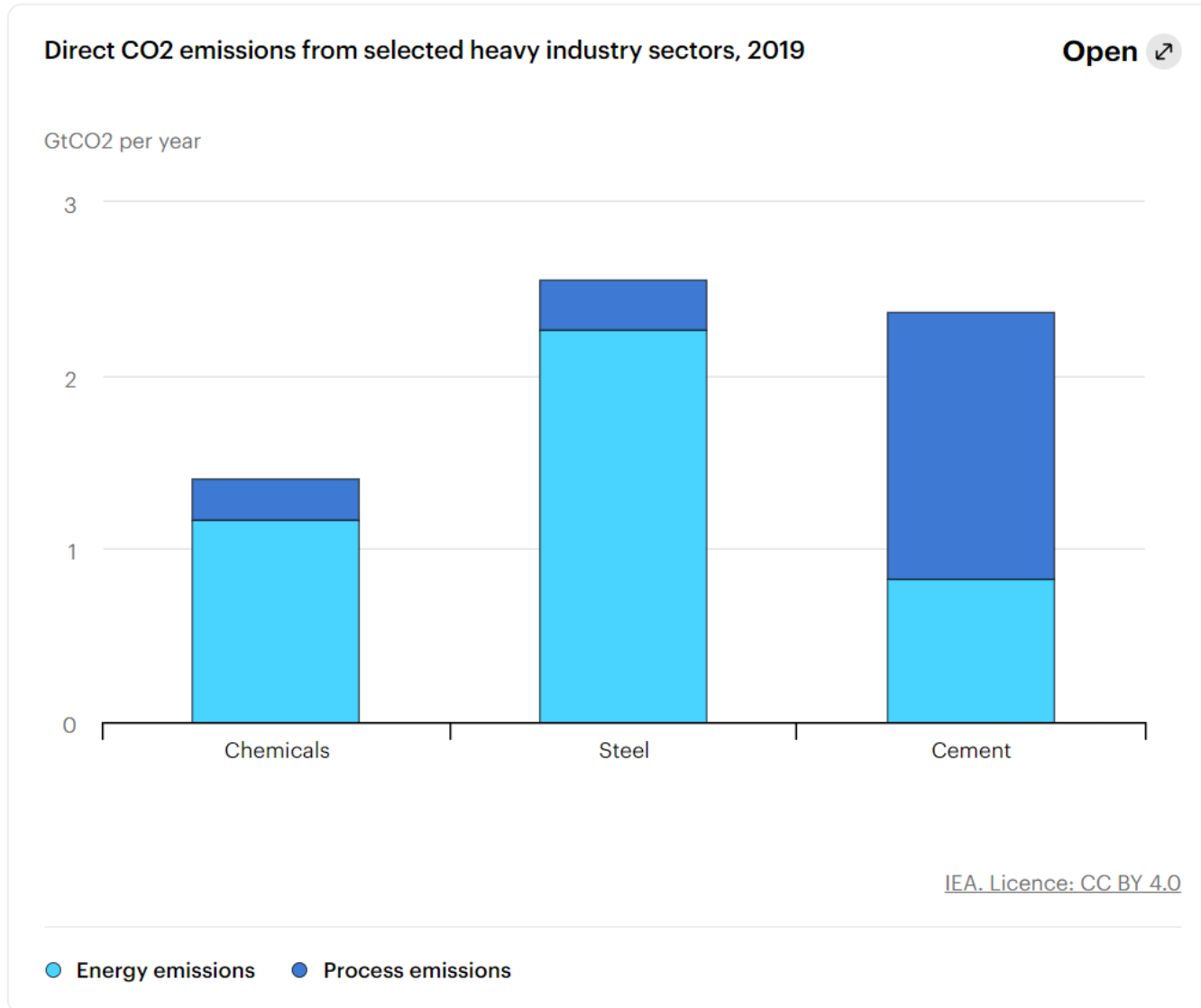


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Coal Oil Gas Electricity Imported heat Bioenergy Other Renewables

- The steel sector is currently the largest industrial consumer of coal, which provides around 75% of its energy demand.
- Coal is used to generate heat and to make coke, which is instrumental in the chemical reactions necessary to produce steel from iron ore.
- *Mtoe?* million tonnes of oil equivalent; unit used to compare energy sources; 1 toe is the amount of energy released by burning 1 tonne of crude oil, about 42 GJ or 12 MWh

[Iron and steel roadmap](#)



- The iron and steel sector directly accounts for 2.6 gigatonnes of carbon dioxide (Gt CO₂) emissions annually
- This is about 7-9% of all anthropogenic CO₂ emissions
- Steel decarbonization strategies are critical!

[Iron and steel roadmap](#)

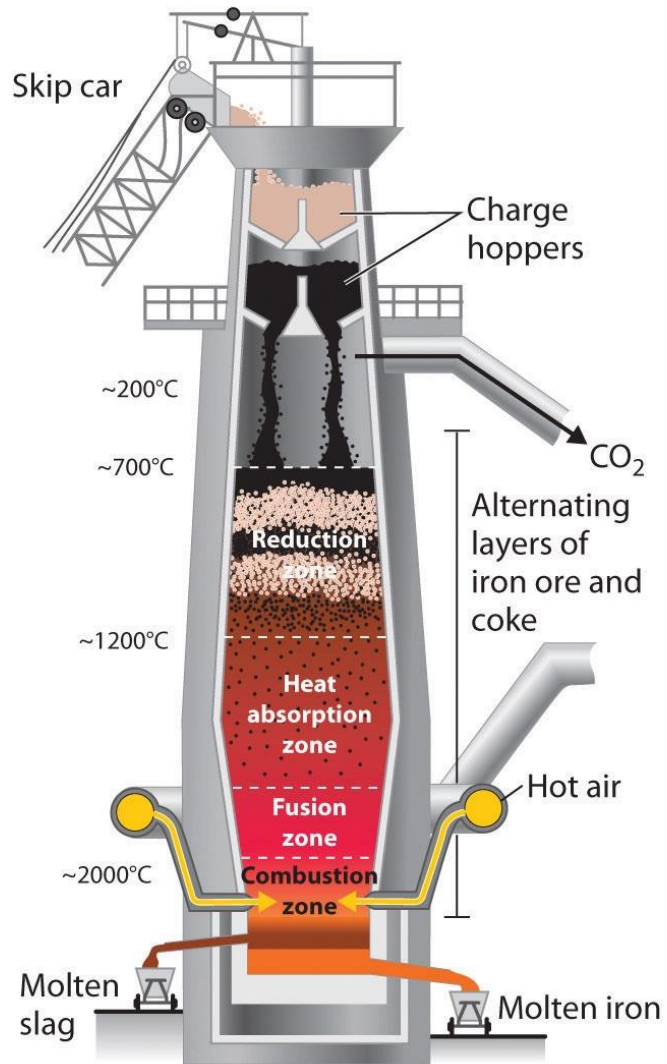
- Starts with resources: mining (ca. 0.25 tCO₂e/t) and recycled
- Next step is smelting to convert iron ore to pig iron, usually in a **blast furnace (BF)**, with hot CO as the reducing agent
- Next step is refining usually in **basic oxygen furnace (BOF)** – oxidation to remove excess carbon and other impurities to get steel
- Alternatively, can use **direct reduced iron (DRI)** instead of blast furnace, with hot H₂ and CO gases as reducing agents + **electric arc furnace (EAF)** to refine (DRI is lower carbon than BF – natural gas or hydrogen alternatives instead of coal)
- EAF instead of BOF (refining alternative)! DRI instead of BF (smelting alternative)!

Steel emissions

Steel production

- BF-BOF emits an average of 2.2 tCO₂e
- DRI+EAF emits an average of 1.2 tCO₂e
- We will briefly explore both approaches

Iron production in a blast furnace



(a) Blast furnace

The furnace is charged with alternating layers of iron ore (largely Fe_2O_3) and a mixture of **coke (C)** and limestone (CaCO_3).

Blasting hot air into the mixture from the bottom causes it to ignite, producing CO and raising the temperature of the lower part of the blast furnace to about 2000 °C.

As the CO that is formed initially rises, it **reduces Fe_2O_3 to form CO_2 and Fe**, which absorbs heat and melts as it falls into the hottest part of the furnace.

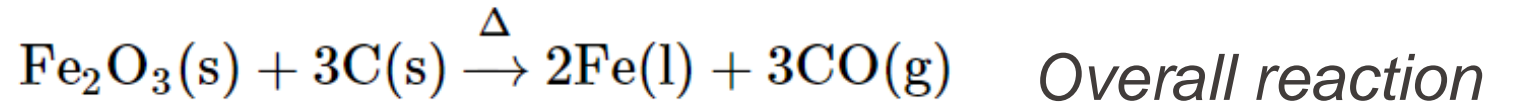
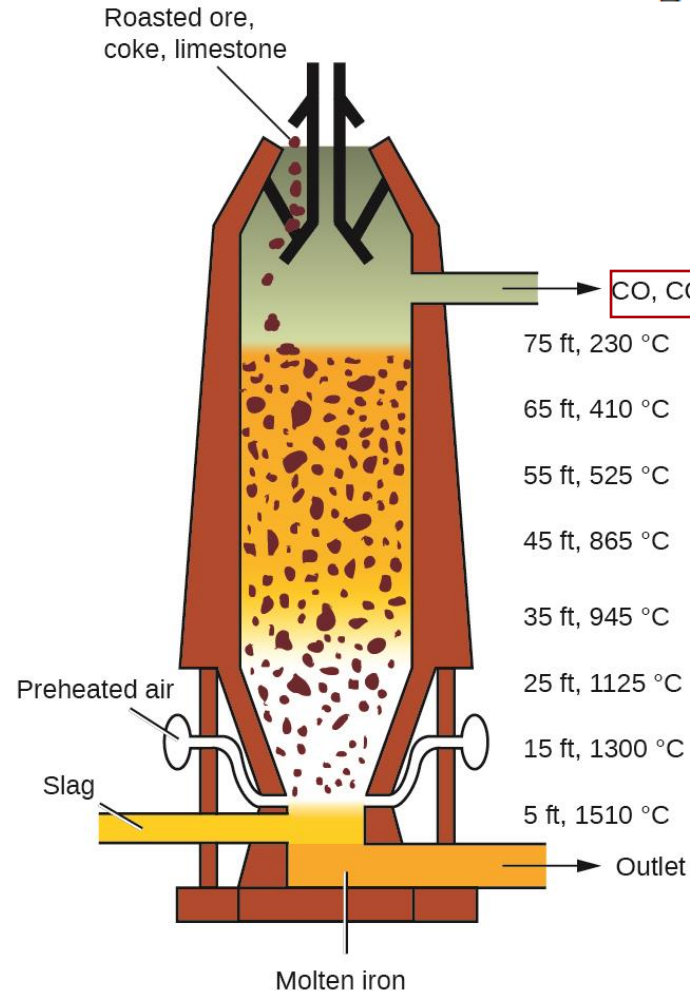
Decomposition of CaCO_3 at high temperatures produces CaO (lime) and additional CO_2 , which reacts with excess coke to form more CO.

Steel making

Smelting = iron ore → pig iron

Iron production making in a blast furnace

1. Furnace is charged up!



CO, CO₂, N₂ 7. Pollution and nitrogen

75 ft, 230 °C



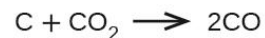
65 ft, 410 °C



55 ft, 525 °C



45 ft, 865 °C



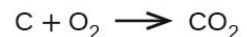
35 ft, 945 °C



25 ft, 1125 °C



15 ft, 1300 °C



5 ft, 1510 °C

6. Slag and Fe are molten, denser slag protects Fe from oxidation; Fe and slag are removed (referred to as **pig iron** – impure)

4. Iron oxides are reduced by CO (reducing agent) generating Fe and CO₂

3. Hot CO₂ is reduced to CO by reaction with coke

5. Limestone decomposes to CaO (lime), which reacts with silicates to form slag (CaSiO₃)

2. Coke (C) reacts with hot air to produce CO₂ and heat

What is the reducing agent that converts iron ore to iron in a blast furnace?

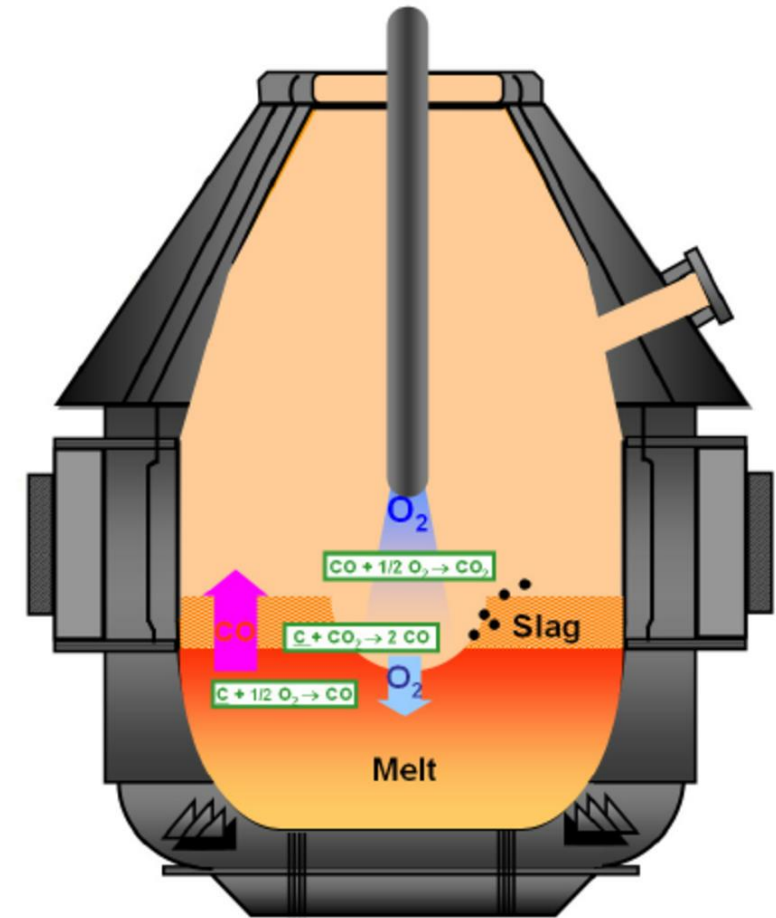
- A. Hot air
- B. Carbon dioxide
- C. Carbon monoxide
- D. Coke

How the molten Fe is converted to steel

- Molten pig iron is refined, usually in a Basic Oxygen Furnace (BOF)

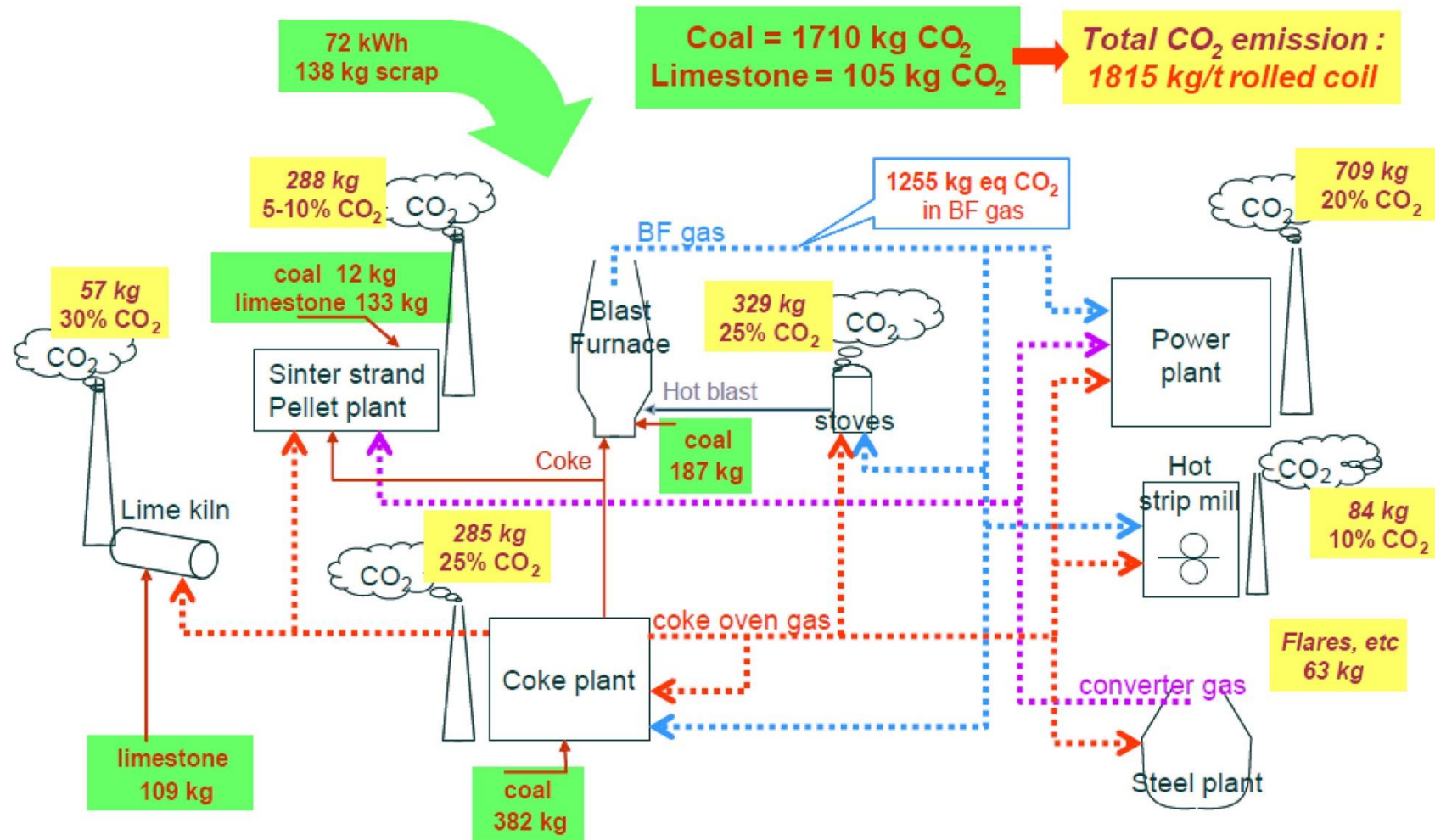
In BOF:

- Molten iron, containing high amounts of carbon, is fed into furnace
- Oxygen blown through at high pressure to oxidize the excess carbon and other impurities to get product known as steel



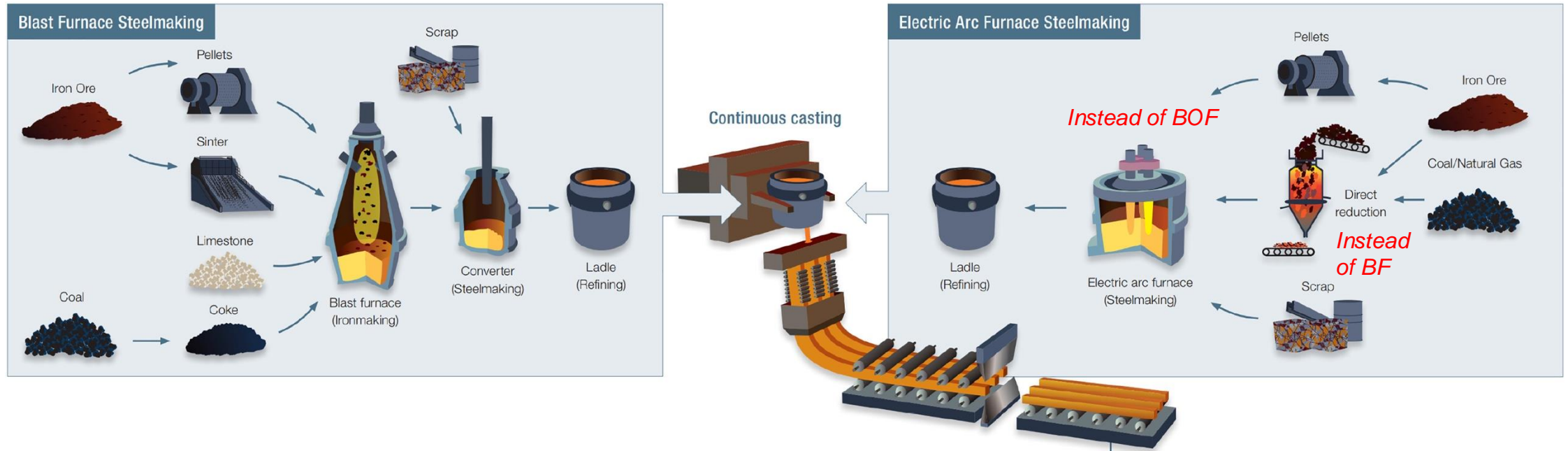
Refining = pig iron \rightarrow steel

Emissions from typical steel mill (blast furnace)



- BF/BOF – blast furnace/basic oxygen furnace
- Primary emission source is raw materials and fuel combustion
- 1.8 t CO₂ emissions per t rolled coil, With 1.7 t from coal use and 0.1 t from lime use*

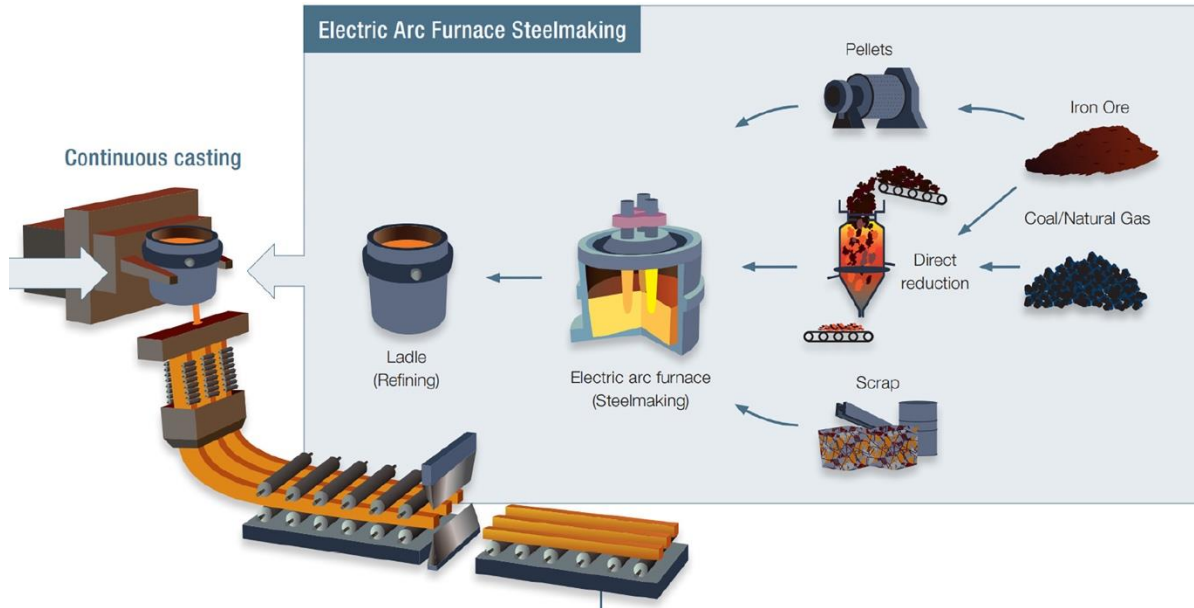
* Some source-related discrepancies



- BF to remove oxygen from ore to get pig iron (smelting)
- BOF to remove excess carbon and remove impurities to get steel
- Coal for energy
- **Best suited for steel from ore**

Decarbonizing steel

- DR removes oxygen from ore to get a usable form of iron (natural gas, syngas, hydrogen for energy)
- EAF is fed with pig iron ("pellets"), DRI, and scrap steel (removal of impurities) – **best suited to steel from scraps**
- EAF uses electricity for energy



- DRI – iron reduced in **solid state**, without melting; lower temperatures ($<1000\text{ }^{\circ}\text{C}$), less carbon, less energy
- In EAF, high voltage arcs are generated, heating and melting scrap steel, DRI and pig iron; less CO_2
- Compared to BF/BOF, more efficient and better for environment (if renewable energy, scrap steel), can be flexible with raw materials (DIR, scrap)

Decarbonizing steel

Two sources of metallics to make steel

Approx. 85-90% of steel scrap potential is recovered

Steel scrap decarbonization

Steel scrap

Secondary sources



“finite”



Iron ore

Primary sources



“expandable”



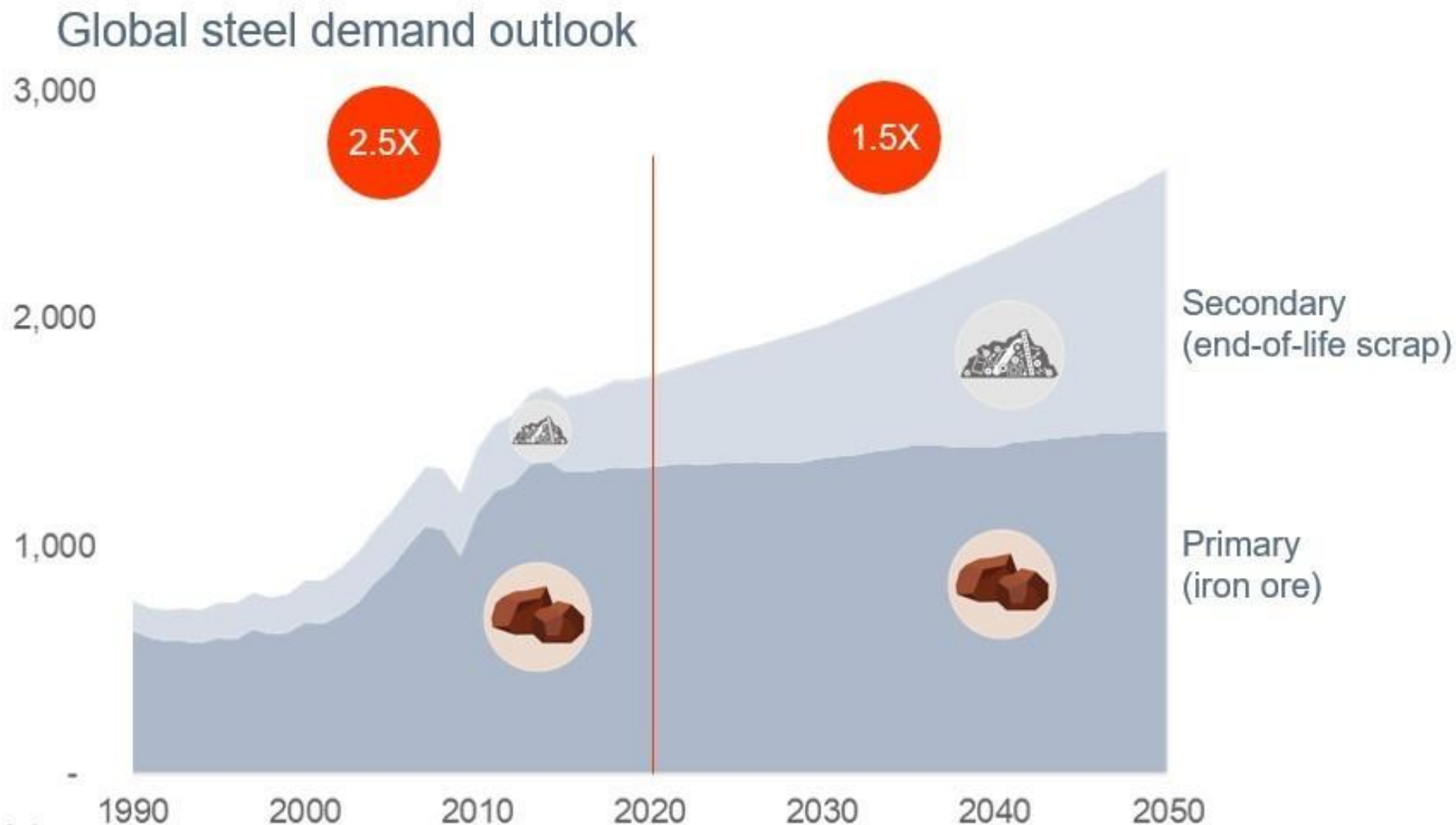
Steel's CO₂ challenge, it's all about primary steel

Steel scrap decarbonization



We cannot escape it, primary steel will be needed over the long term

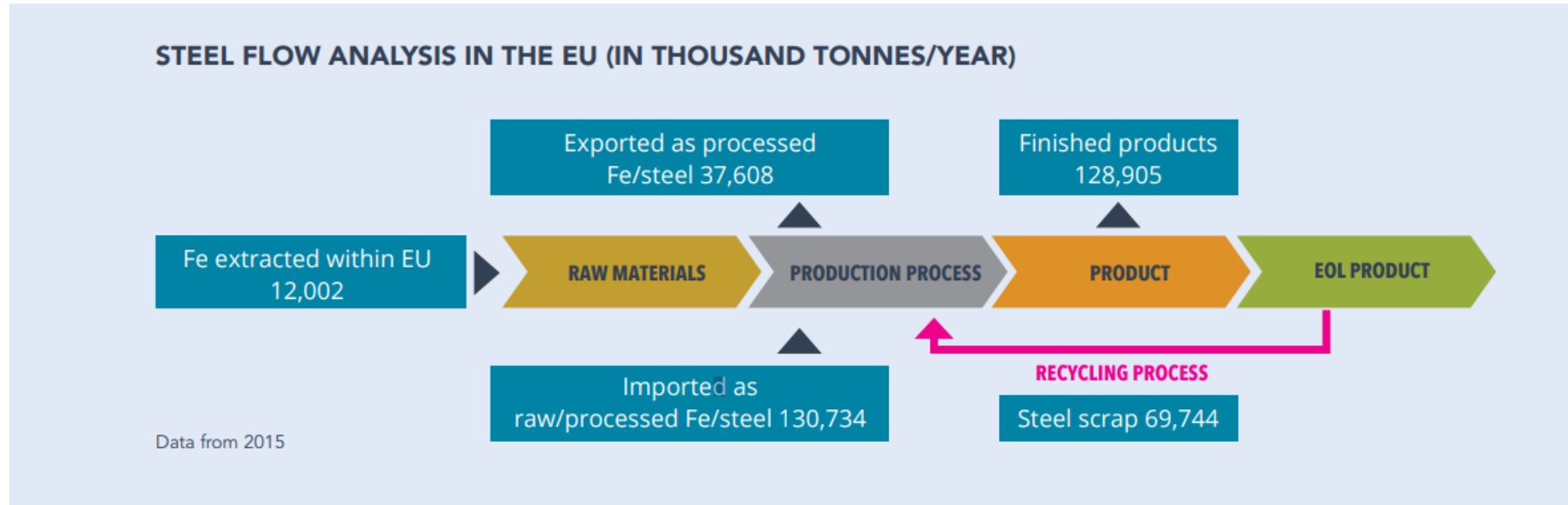
Steel scrap decarbonization



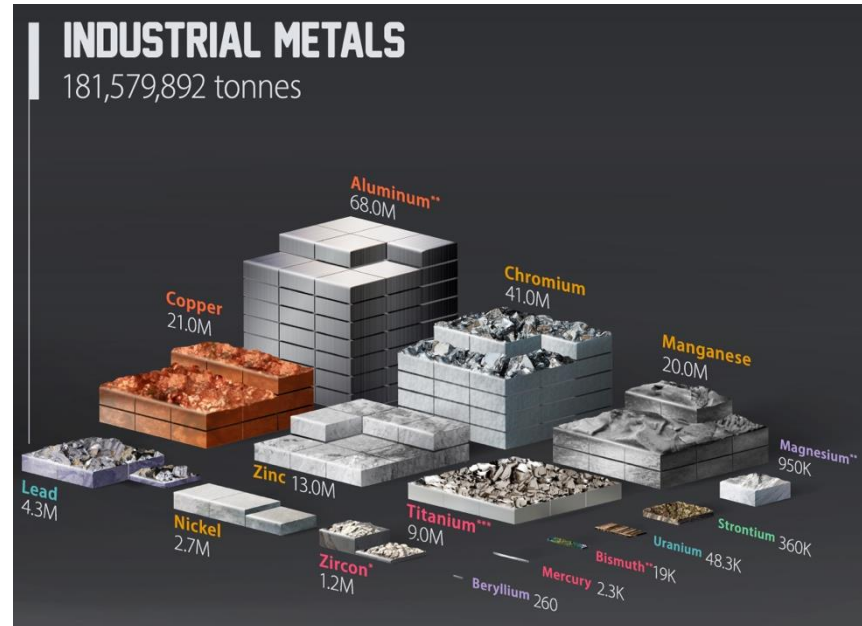
Key drivers of steel demand growth

- Developing world convergence to developed world standards
- Transition to clean energy economy
- Partially neutralised by material efficiency

Some strategies	Characteristics
Improve energy efficiency	<ul style="list-style-type: none">• Exactly as it sounds
Reduction with hydrogen	<ul style="list-style-type: none">• Instead of carbon-intensive coke (coal derived) used hydrogen as a reducing agent
Carbon capture, utilization and storage	<ul style="list-style-type: none">• Exactly as it sounds (will discuss in a later class)
Material efficiency and recycling	<ul style="list-style-type: none">• Maximize the use of scrap steel, design more efficiently to use less steel
Biomass	<ul style="list-style-type: none">• Replace fossil fuels with biomass

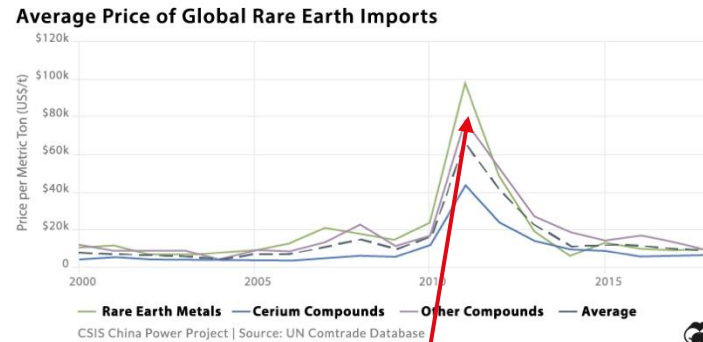
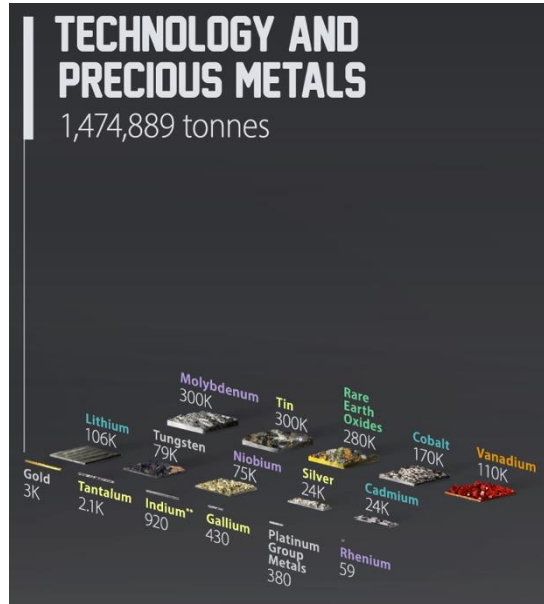


- Using steel scrap reduces CO₂ emissions by 58%.
- Recycling steel saves 72% of the energy needed for primary production.
- Recycling one tonne of steel saves 1.4 tonnes of iron ore, 0.8 tonnes of coal, 0.3 tonnes of limestone and additives, and 1.67 tonnes of CO₂.
- In 2018, 157 million tonnes of CO₂ were saved in the EU by recycling 94 million tonnes of scrap, an equivalent amount to all automobiles circulating in France, Great Britain and Belgium.
- Using recycled steel to make new steel reduces air pollution by 86%, water use by 40%, and water pollution by 76%.



- **Aluminum**, most abundant metal in the Earth's crust (8% by weight)
 - Around 75% of the 900 million tons ever produced is still in use
 - Found in electronic devices, vehicles, and aircrafts
- **Copper**, used in many applications ranging from wiring to construction due to its durability, malleability, and electrical and thermal conductivity
- **Manganese**, fundamental in steel production and lithium-ion batteries

Rare earth metals



2010, China reduces REM export by 37%

- **Rare Earth Metals**, critical for current high-tech devices (e.g., screens, catalysts, batteries)
- **Lithium and Cobalt**, fundamental for lithium-ion batteries (e.g., electric vehicles)

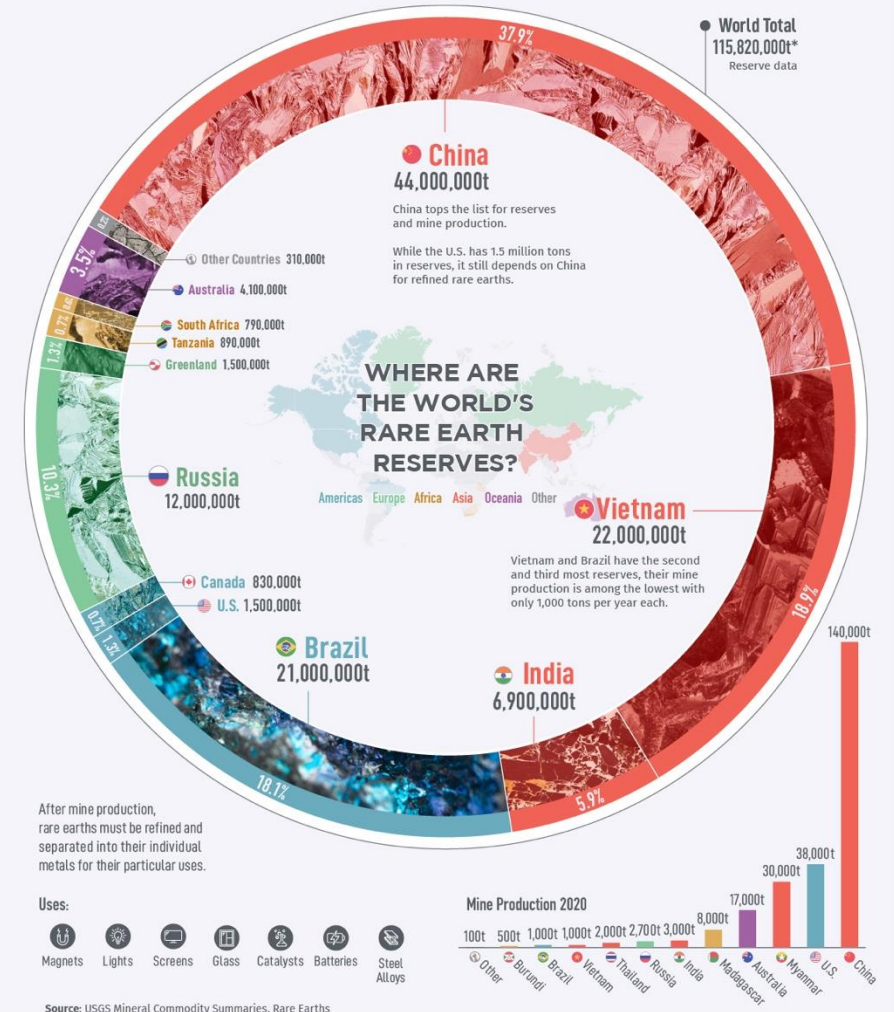


WHERE IN THE WORLD ARE ALL THE RARE EARTHS?

Rare earth elements (REEs) are a group of 17 elements whose importance is critical in high technology. Their use has exploded as electronics and renewable technologies increasingly have become part of everyone's daily lives.

Rare earths are abundant in the Earth's crust but mineable concentrations are less common, making reserves potential very valuable and strategic.

*The USGS tracked the world's reserves in tons (imperial).



Periodic Table of the Elements

IA																VIIIA																			
1 H Hydrogen																2 He Helium																			
3 Li Lithium	4 Be Beryllium																5 B Boron	6 C Carbon	7 N Nitrogen	8 O Oxygen	9 F Fluorine	10 Ne Neon													
11 Na Sodium	12 Mg Magnesium																13 Al Aluminium	14 Si Silicon	15 P Phosphorous	16 S Sulphur	17 Cl Chlorine	18 Ar Argon													
19 K Potassium	20 Ca Calcium	21 Sc Scandium	22 Ti Titanium	23 V Vanadium	24 Cr Chromium	25 Mn Manganese	26 Fe Iron	27 Co Cobalt	28 Ni Nickel	29 Cu Copper	30 Zn Zinc	31 Ga Gallium	32 Ge Germanium	33 As Arsenic	34 Se Selenium	35 Br Bromine	36 Kr Krypton																		
37 Rb Rubidium	38 Sr Strontium	39 Y Yttrium	40 Zr Zirconium	41 Nb Niobium	42 Mo Molybdenum	43 Tc Technetium	44 Ru Ruthenium	45 Rh Rhodium	46 Pd Palladium	47 Ag Silver	48 Cd Cadmium	49 In Indium	50 Sn Tin	51 Sb Antimony	52 Te Tellurium	53 I Iodine	54 Xe Xenon																		
55 Cs Caesium	56 Ba Barium	57-71 Lan.	72 Hf Hafnium	73 Ta Tantalum	74 W Tungsten	75 Re Rhenium	76 Os Osmium	77 Ir Iridium	78 Pt Platinum	79 Au Gold	80 Hg Mercury	81 Tl Thallium	82 Pb Lead	83 Bi Bismuth	84 Po Polonium	85 At Astatine	86 Rn Radon																		
87 Fr Francium	88 Ra Radium	89-103 Act.	104 Rf Rutherfordium	105 Db Dubnium	106 Sg Seaborgium	107 Bh Bohrium	108 Hs Hassium	109 Mt Meitnerium	110 Ds Darmstadtium	111 Rg Roentgenium	112 Cn Copernicium	113 Nh Nihonium	114 Fl Flerovium	115 Mc Moscovium	116 Lv Livermorium	117 Ts Tennesine	118 Og Oganesson																		
Lanthanides			57 La Lanthanum	58 Ce Cerium	59 Pr Praseodymium	60 Nd Neodymium	61 Pm Promethium	62 Sm Samarium	63 Eu Europium	64 Gd Gadolinium	65 Tb Terbium	66 Dy Dysprosium	67 Ho Holmium	68 Er Erbium	69 Tm Thulium	70 Yb Ytterbium	71 Lu Lutetium																		
Actinides			89 Ac Actinium	90 Th Thorium	91 Pa Protactinium	92 U Uranium	93 Np Neptunium	94 Pu Plutonium	95 Am Americium	96 Cm Curium	97 Bk Berkelium	98 Cf Californium	99 Es Einsteinium	100 Fm Fermium	101 Md Mendelevium	102 No Nobelium	103 Lr Lawrencium																		

ELEMENT GROUPS

Non Metals

Halogens

Noble Gases

Metals

Metalloids

Alkali Metals

Alkali Earth Metals

Transition Metals

Lanthanides

Actinides

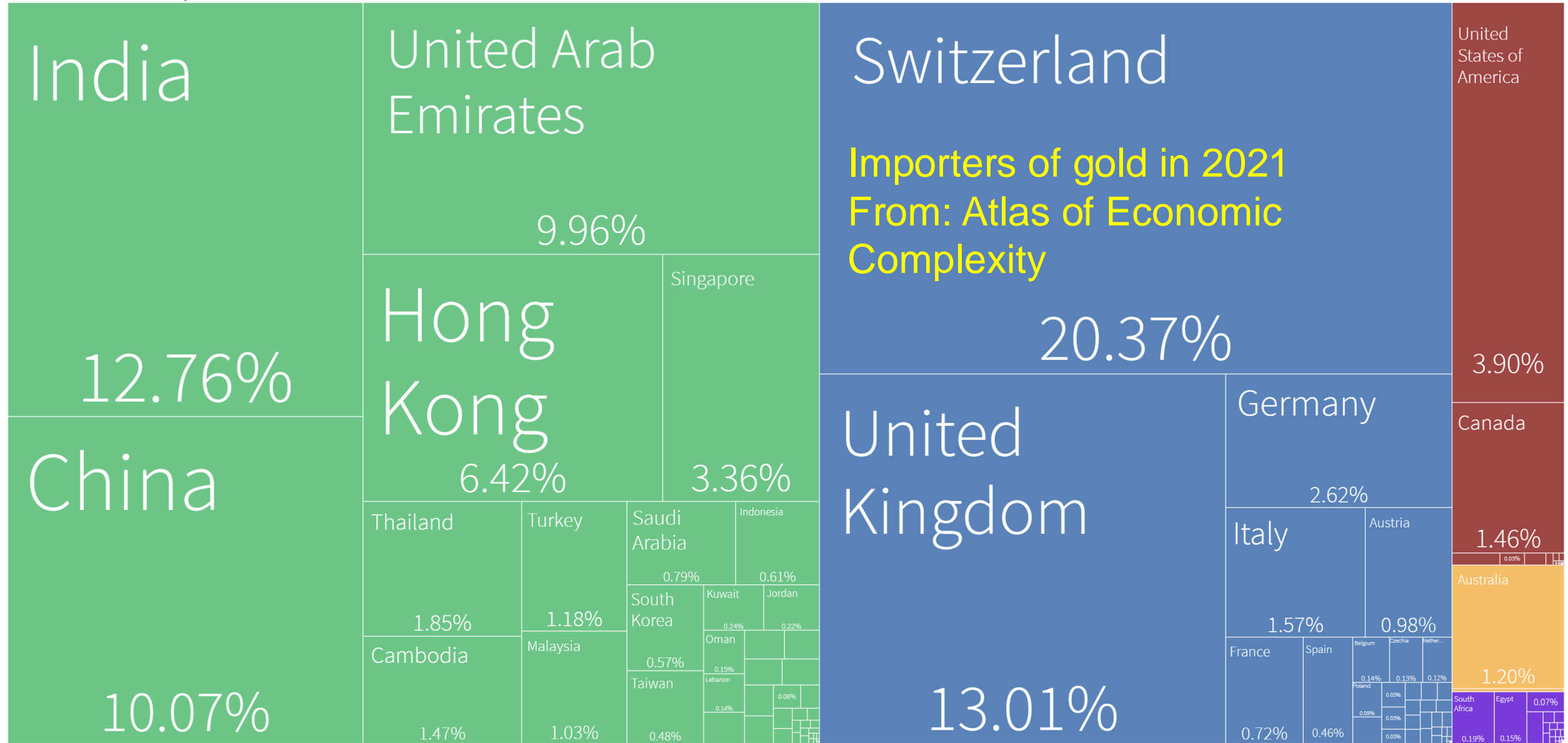
How many rare earth elements in an iPhone?

- A. 3
- B. 6
- C. 8
- D. 17



- 4/7 global gold refineries their Switzerland
- **Watches, jewelry, and investments** are by far the main uses of gold, and Switzerland is a world leader in all 3 areas
- Demand for new mining operations – due to growth in investment sector
- Lack of transparency in sourcing – most watch companies were not able to trace their supply chain to understand where their raw materials were sourced (2018 WWF study)





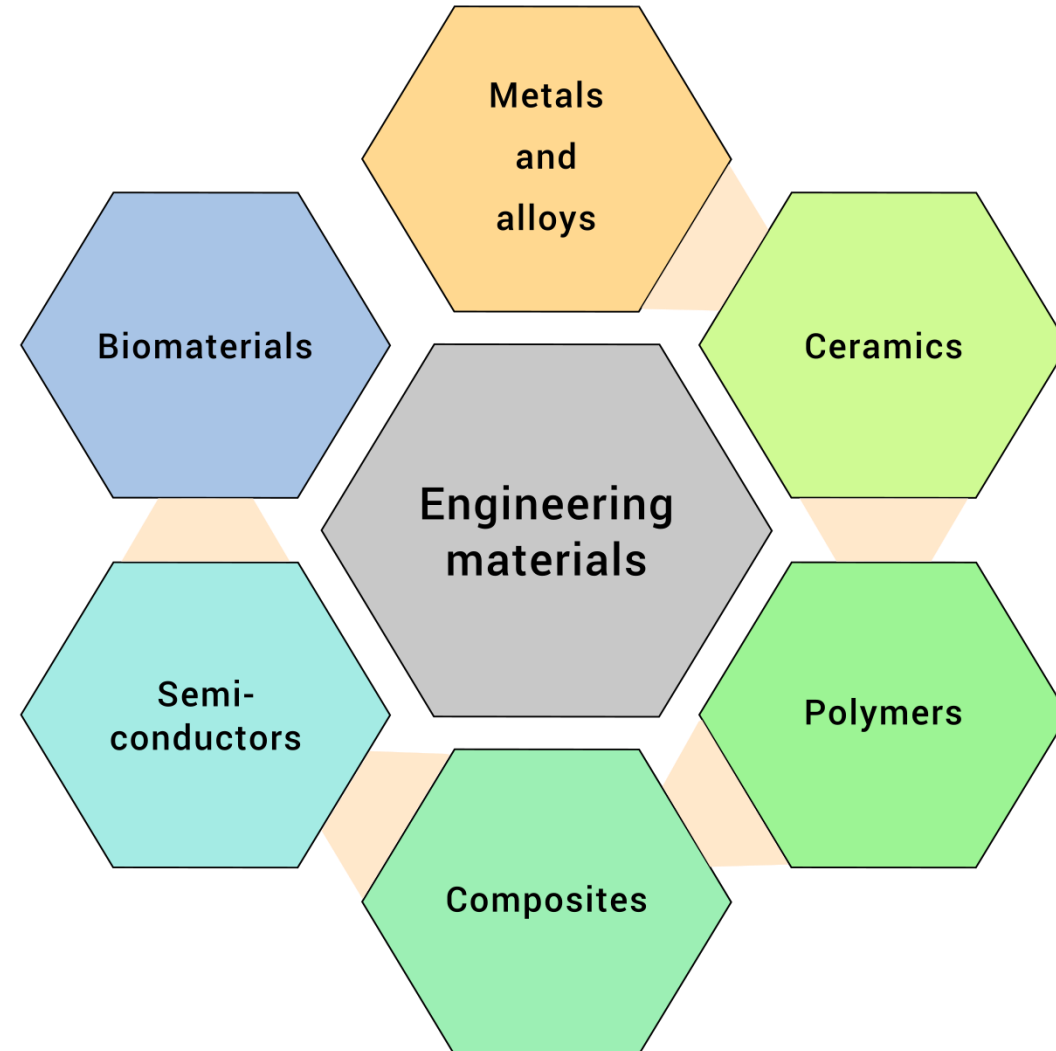
Gold stats – executive summary

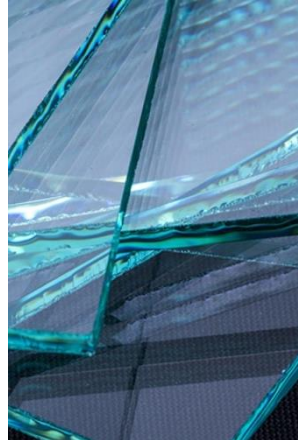
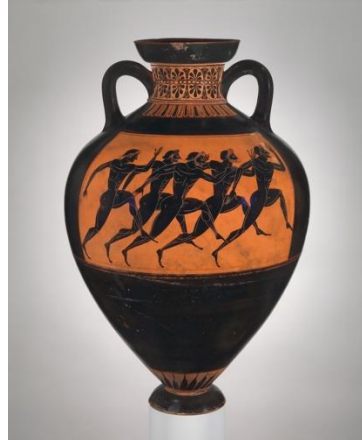
- 1 tonne of gold produces 100 000 tonnes of waste rock (or 1000 kg of soil moved for one 10 g ring!)
- 1 kg of gold generates 12 500 kg of CO₂ equivalents (almost 3× Switzerland's transportation emission)
- 38% of worldwide total anthropogenic mercury emissions in 2015 were attributed to artisanal and small-scale mining (focus in Sub-Saharan Africa and South America – 163% increase between 2010-2015)
- Large scale mining (80% of global gold mine production) also challenging with local environmental problems as massive deposits are mined after long periods of time – contaminants, heavy metals, acid mine drainage...

From: The impact of gold Sustainability aspects in the gold supply-chains and Switzerland's role as a gold hub – 2021 WWF report

Engineering materials categories (variable)

- Metals and alloys
- Polymers
- **Ceramics**
- Composites
- Semiconductors
- Biomaterials





- Ceramics are crystalline, which contributes their almost always high melting temperatures and thermal stability
- Glasses are amorphous and soften/melt upon heating

[illegible]

- Considered one of the greatest and earliest successes of humankind – control of fire
- Among the 1st objects manufactured
- From Greek “keramos” meaning burned earth
- Defined as non-metallic inorganic solids, obtained by firing powders
- Characteristics – long service life, low density, chemically inert, corrosion resistant, electromagnetic response, non-toxic, heat and fire resistant, sometimes electrical resistance or porosity

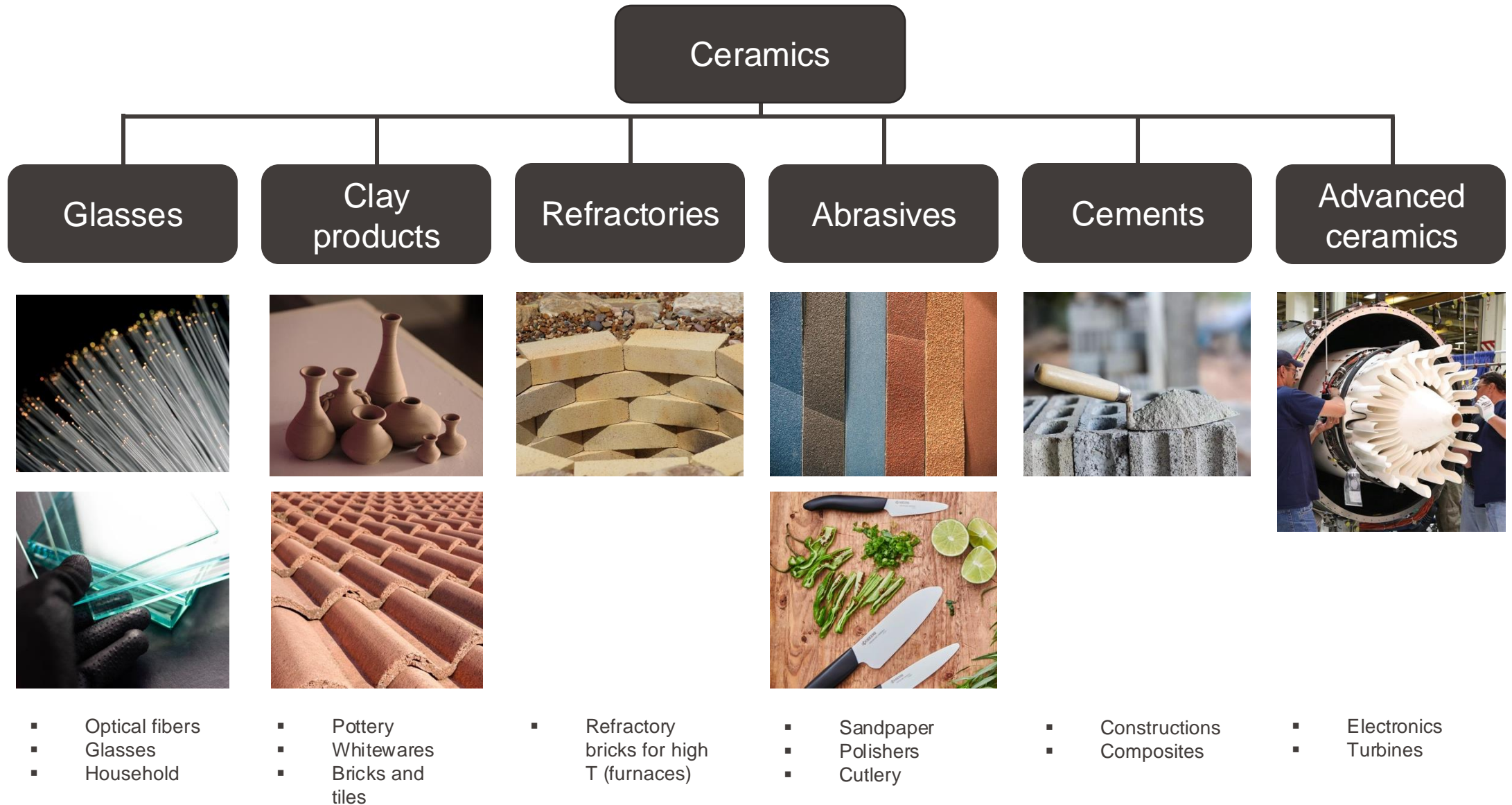


[Decarbonizing ceramics 2022](#)

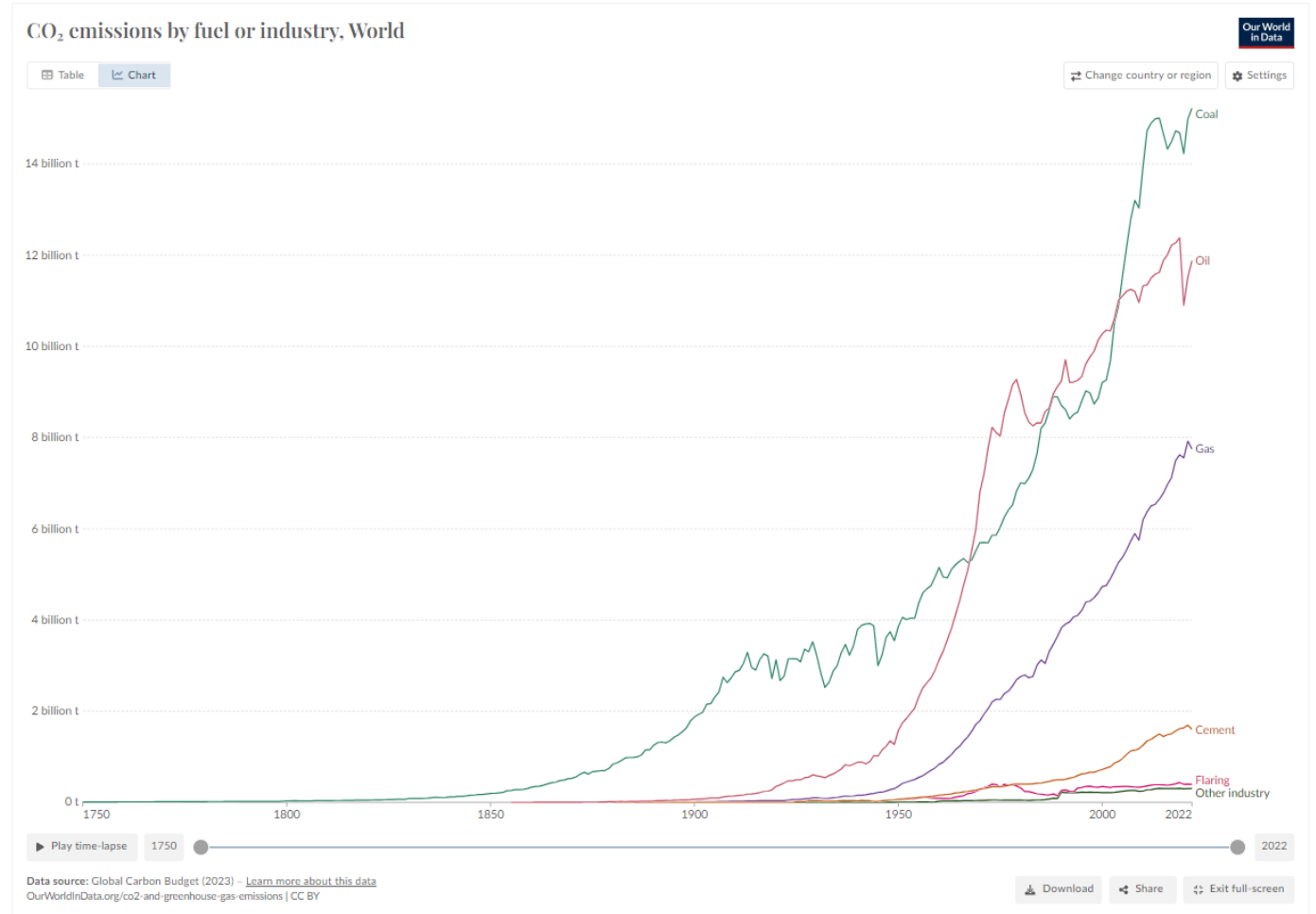
- 1st pieces reported ca. 24,000 years ago
- Many uses today: promising for aerospace and high temperature structural applications, information storage and optical devices, oral prosthetics, water purification, bone void fillers, CO₂ adsorbents, etc.,
- Value of global ceramics market was approx. 230 billion in 2018, and growing due to constant growth in the construction industry, technological advancements in nanotechnology, 3D printing, and ceramics in health



Main uses – Ceramics and Glasses



- In EU: production of refractories, wall and floor tiles, and bricks and roof tiles emit around 19 Mt CO₂
- Globally: brick manufacture is responsible for 2.7% of annual carbon emission; cement (considered a ceramic in this course's context) is responsible for 5-8% of annual emissions



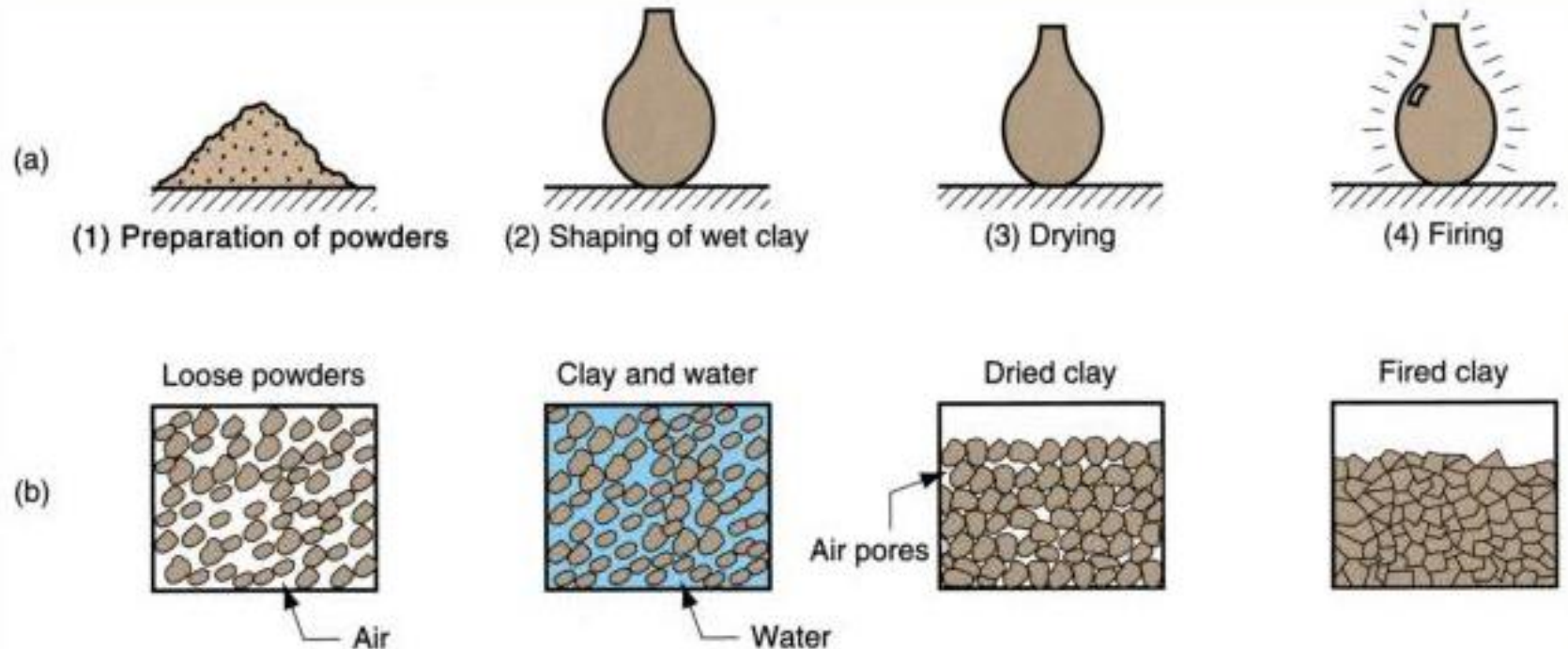
[Decarbonizing ceramics 2022](#)

[Projecting future carbon emissions from cement 2023](#)

Traditional ceramics

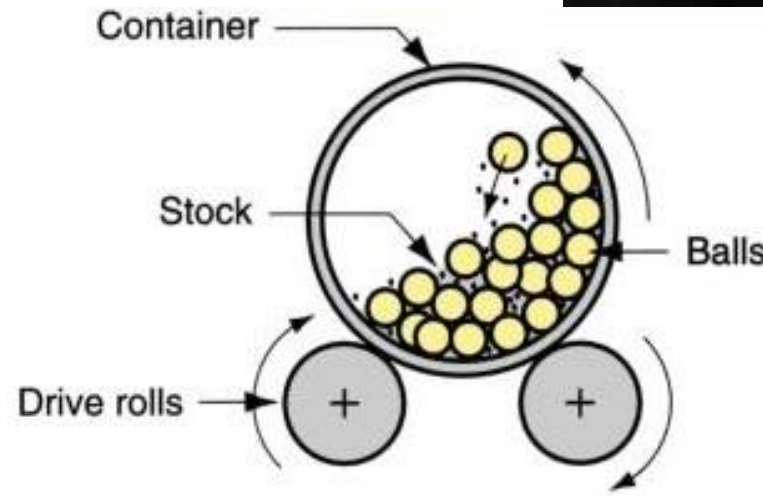
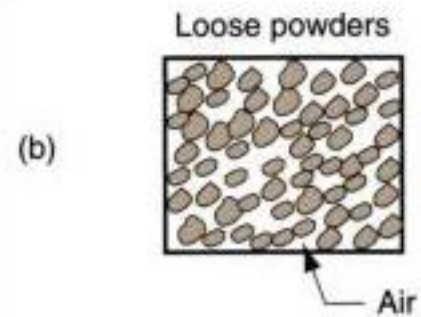
Processing ceramics

- Traditional ceramics are made from naturally occurring minerals
- Mineral extraction has the usual consequences: habitat destruction, soil erosion, water pollution
- Open pits and quarries common

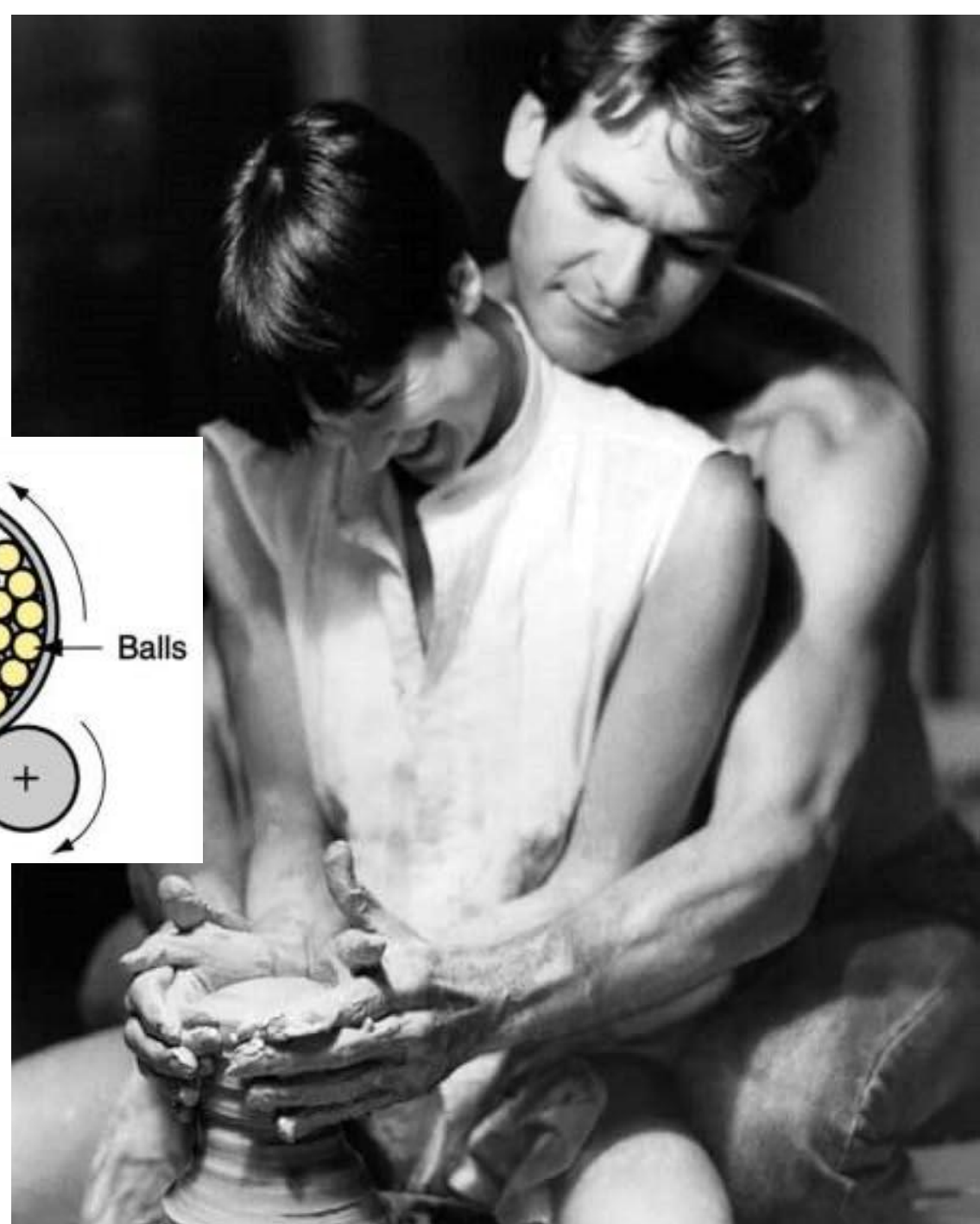


Powders

Processing ceramics

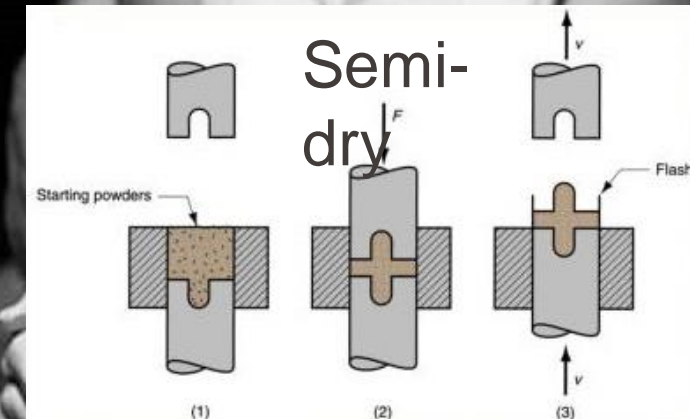
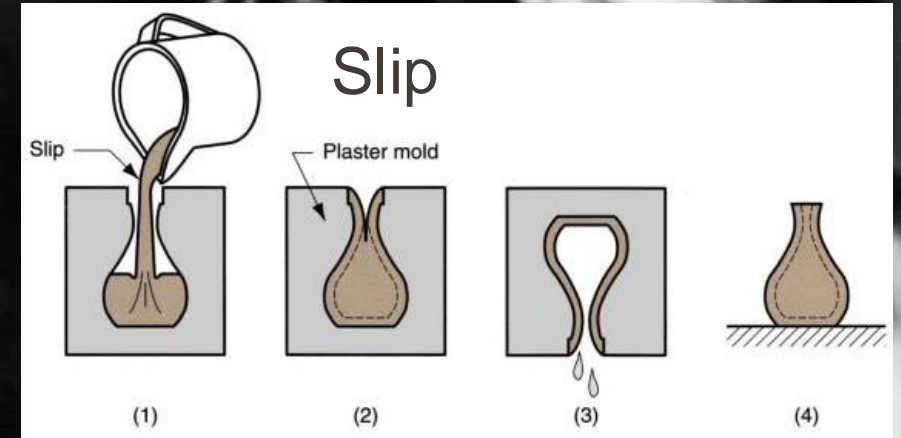
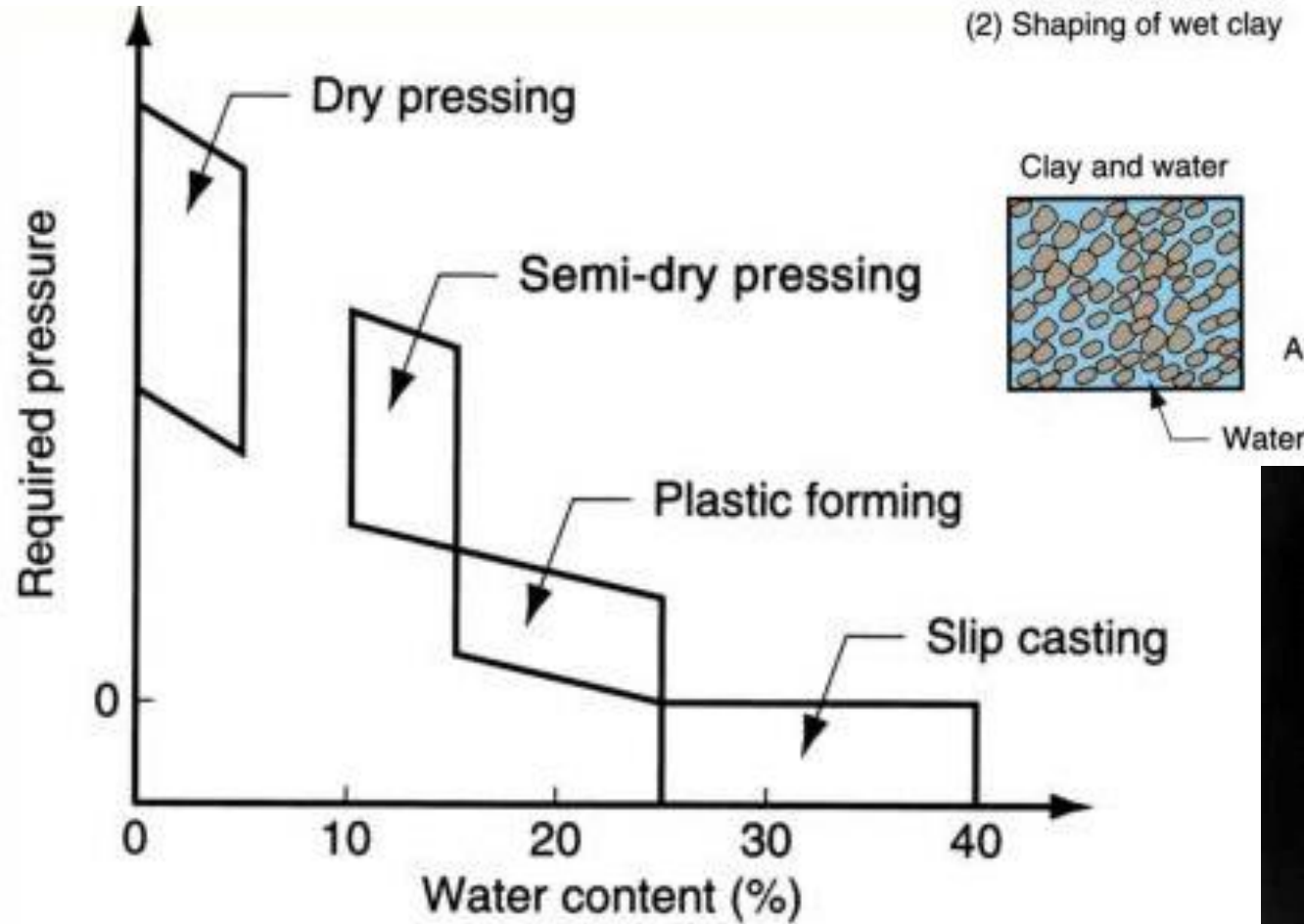


- Starts with a powder
- Powder usually made by ball crushing and then grinding, which is very energy intensive!



Shaping

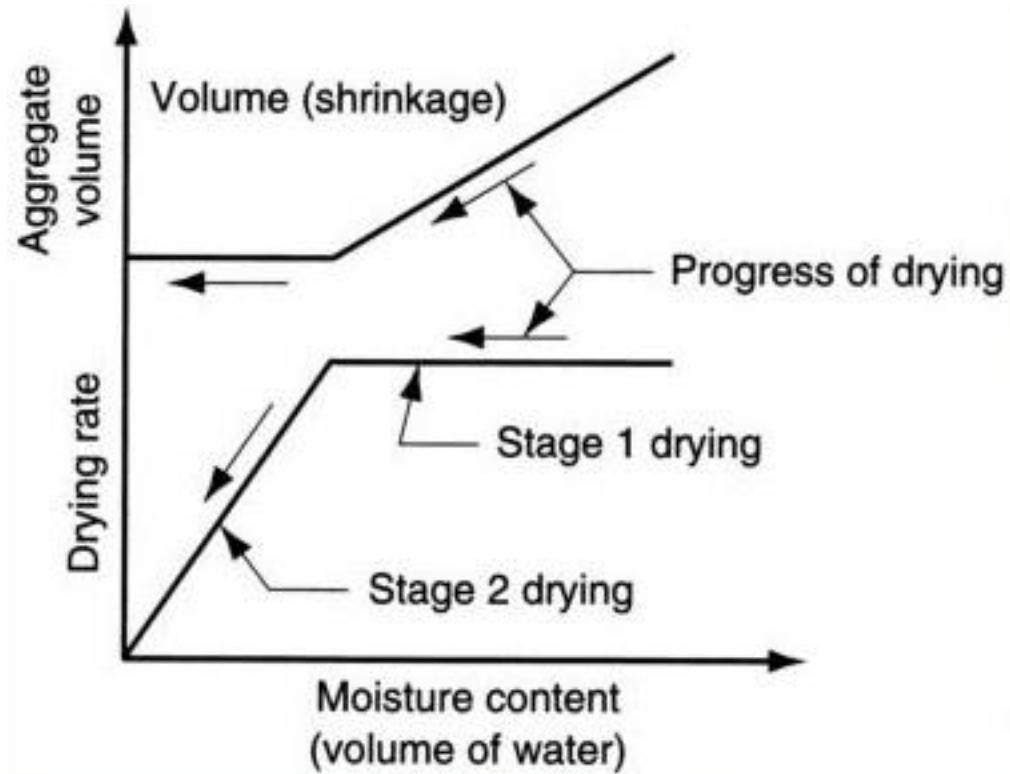
Processing ceramics



- Water is needed for shaping only/plasticity
- High pressures required for semi-dry/dry shaping

Drying – 2 stages

Processing ceramics



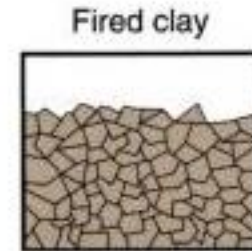
- Stage 1: fast and constant rate to evaporate surface water (prone to shrinkage)
- Stage 2: enough water has been removed so particles are in contact (no shrinkage)



Firing

Processing ceramics

- Heat treatment to *sinter* – form bonds between particles, occurring with densification
- Performed in kiln
- Biggest impact to emissions, related to use of fossil fuels to heat the kiln and process emissions from carbonate raw materials which emit CO_2 when heated
- If glaze is applied, the ceramic is fired again!





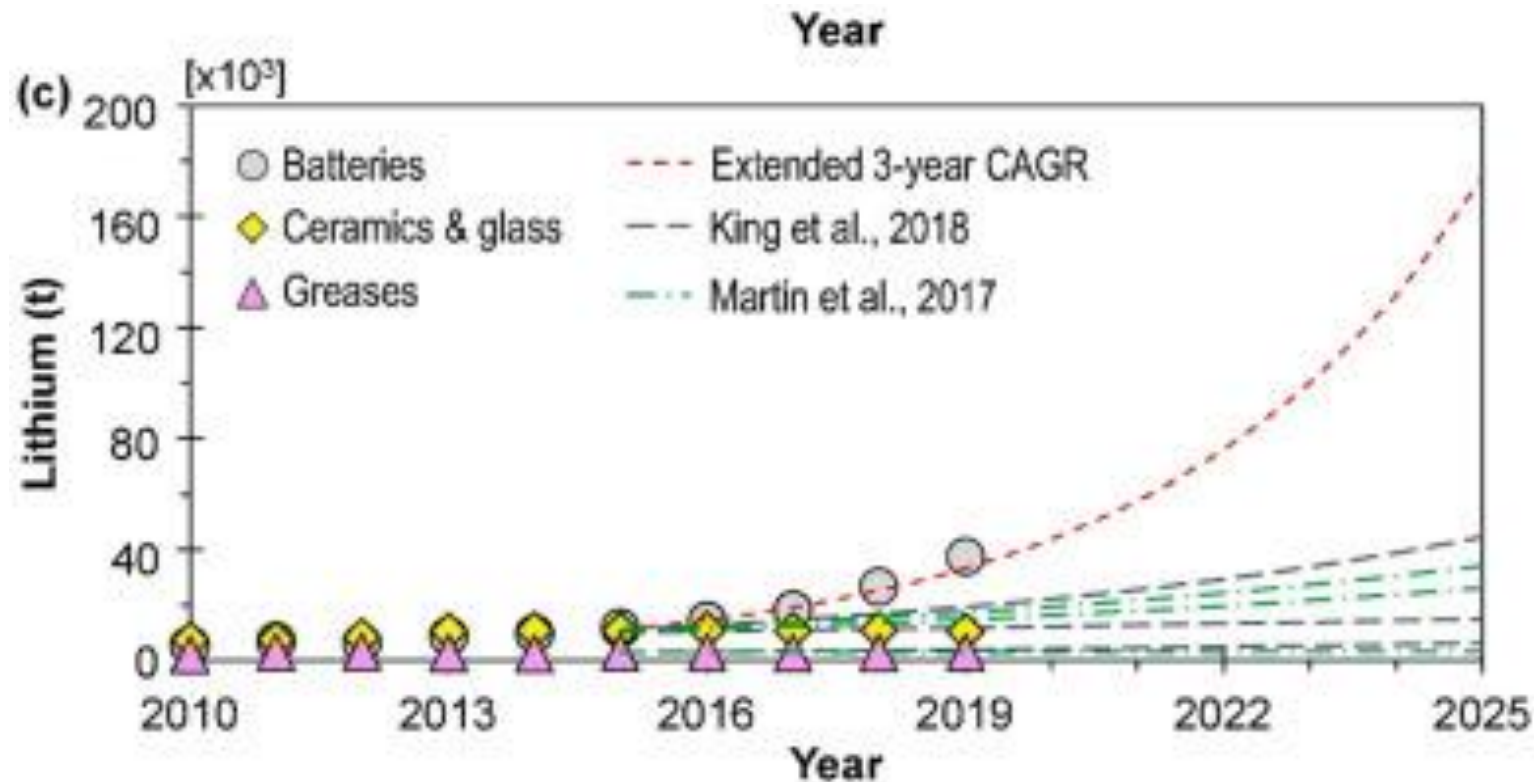
- Made from manmade raw materials
- Developed to meet industrial, technological, and biomedical needs
- Exhibit exceptional properties: high strength, resistance to wear, electrical insulation or conductivity, piezoelectricity, etc.,

Same general steps:

1. **Starting material preparation** – to satisfy high strength requirements grain size is smaller! Controlled by mechanical and chemical methods
2. **Shaping** – some methods borrowed from powder metallurgy, like hot pressing, which sinters as you press; separate firing step is not needed, or injection molding, where ceramic particles are carried by a polymer that is removed after shaping; lower temperatures need to shape but sintering still needed
3. **Sintering** – plasticity is not usually based on water so drying is not needed, but sintering is still needed
4. **Finishing** – to improve finish or dimensions, usually abrasive



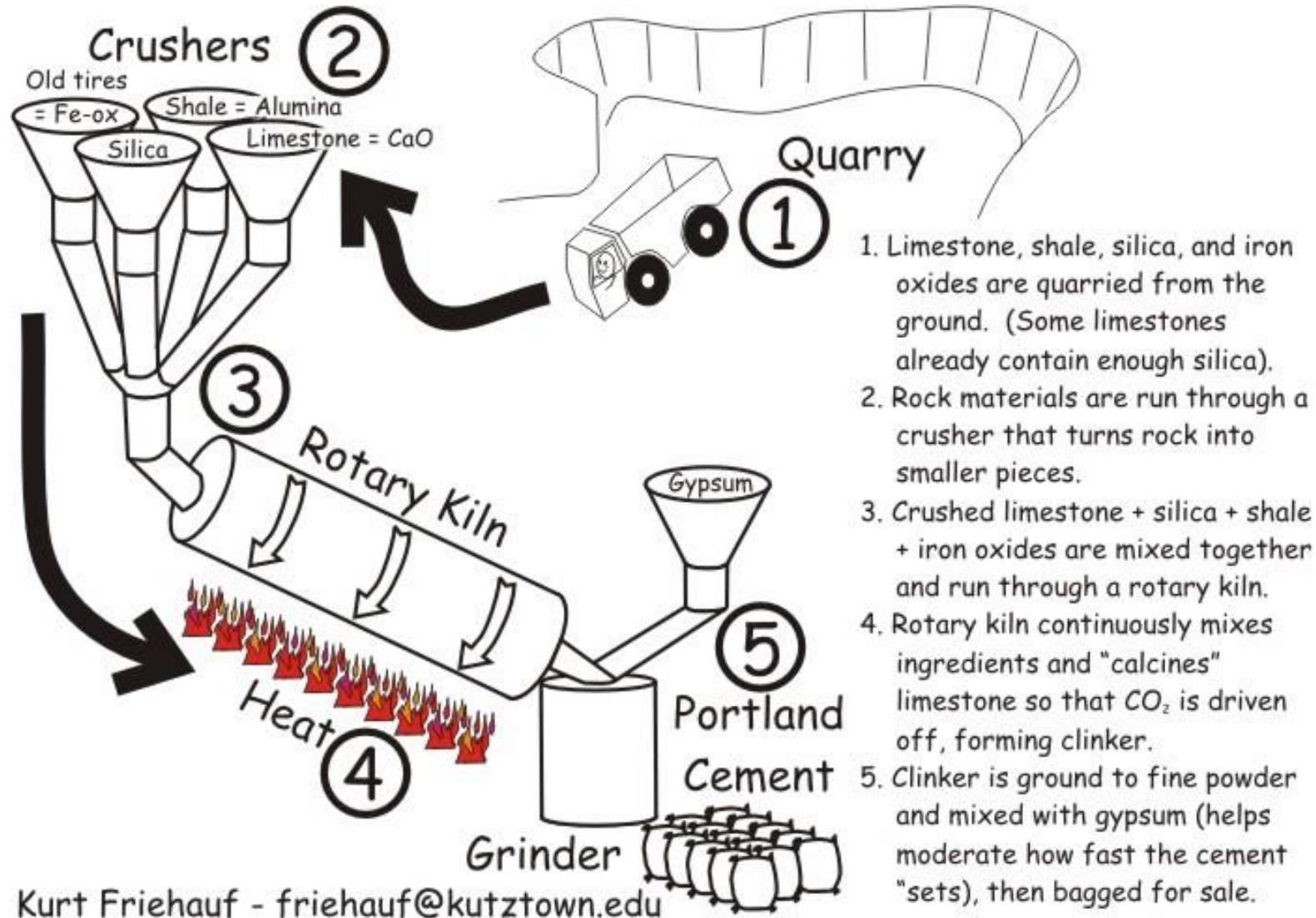
- Lithium is used in new ceramics due to its ability to enhance certain properties: low thermal expansion, high strength and toughness, melting point reduction, improved optical properties, enhanced electrical properties, lightweight



- Now batteries, but before 2015, ceramics and glass was the main industry using lithium

[Decarbonizing ceramics 2022](#)

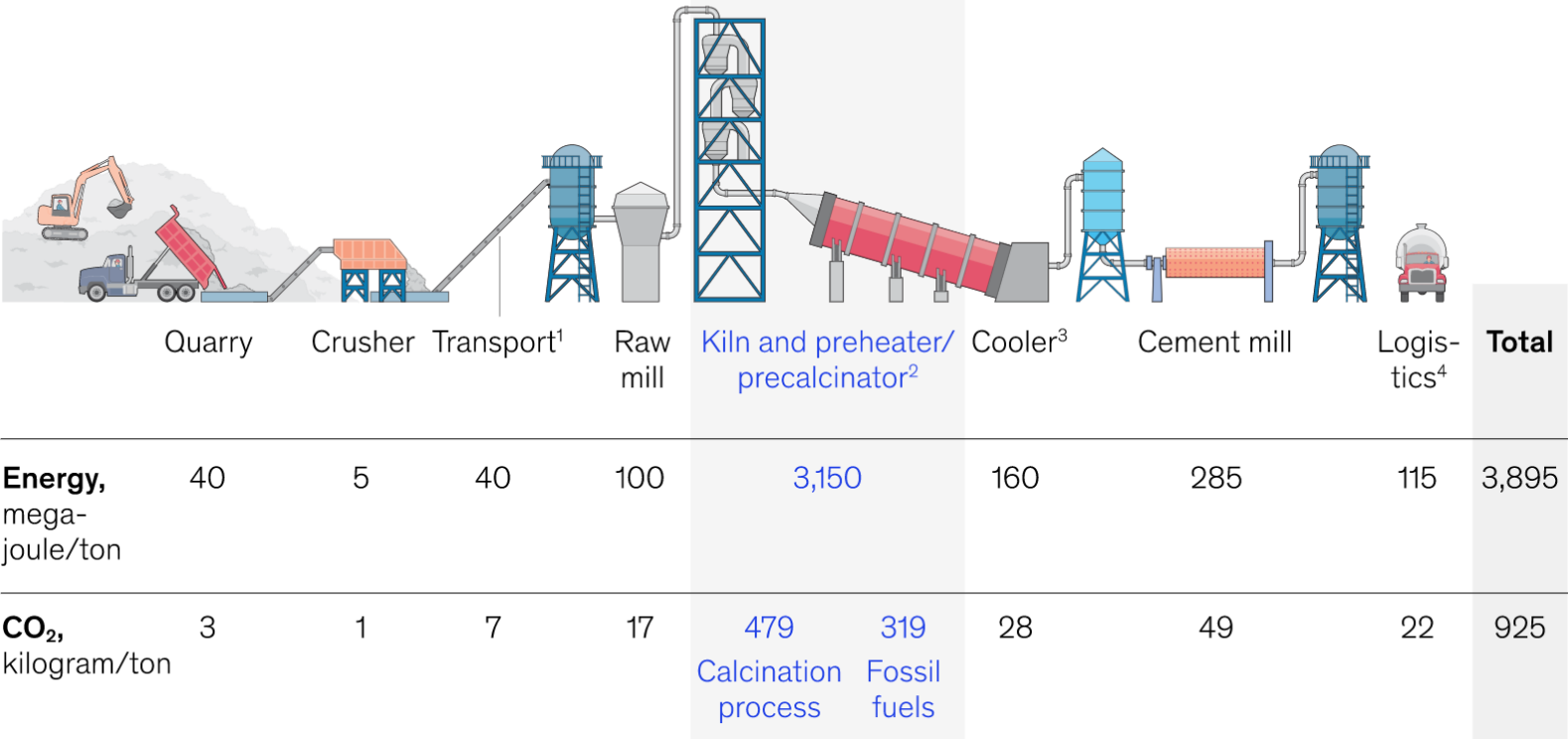
How cement is made



Cement manufacture

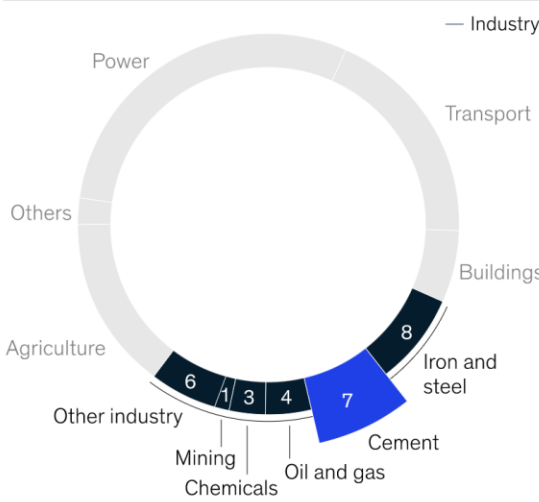
Raw materials, energy, and resources

Clinker and cement manufacturing

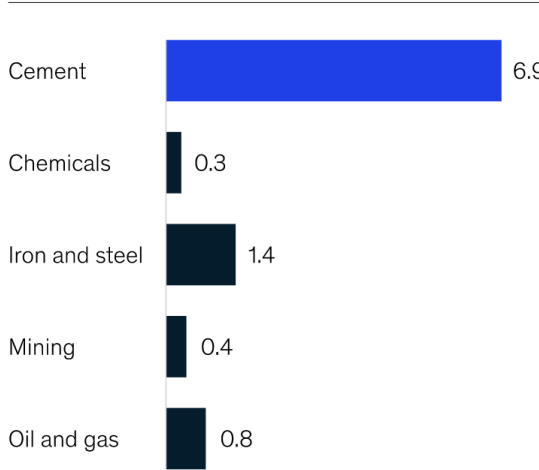


- The cement industry alone is responsible for about a quarter of all industry CO₂ emissions, and it also generates the most CO₂ emissions per dollar of revenue
- About two-thirds of those total emissions result from calcination, the chemical reaction that occurs when raw materials such as limestone are exposed to high temperatures.

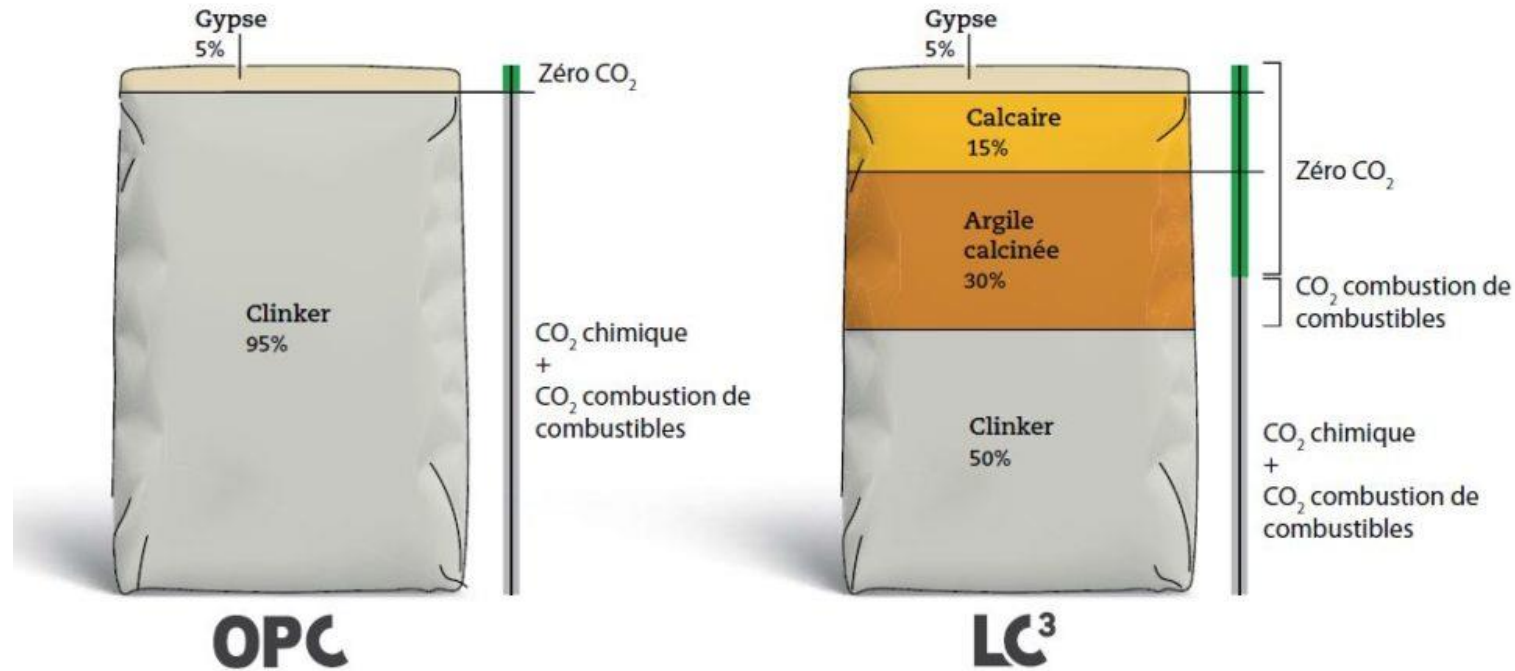
Share of global CO₂ emissions, % in 2017



kg of CO₂ per \$

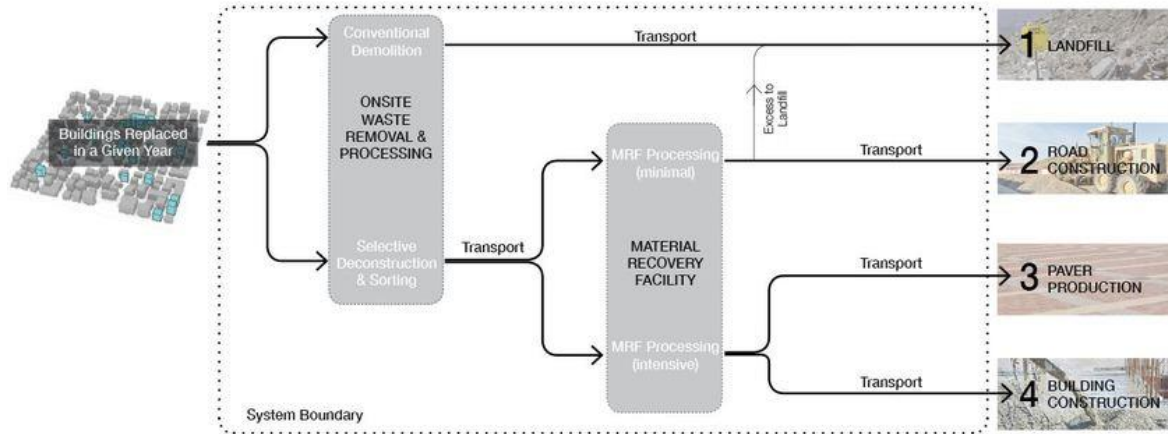


Decarbonizing cement : EPFL connection



- Replaces half of clinker with calcined clay and ground limestone, neither of which releases CO₂ when heated the way limestone does
- Clay is heated to lower temperatures, reducing fuel needed and emissions
- Lower temperatures mean that cleaner electricity like clinker can be used

End of life – Ceramics and Glasses



Concrete

- About 60% of what is crushed can be used for downcycling processes (recovery of materials with a more limited range of uses than the original material).
- These fragments can be used as base materials for structures such as roads.



<https://www.archdaily.com/933616/is-it-possible-to-recycle-concrete>

Moving on:

- ~~Snapshot of the Swiss economy (recall, less than 7% circular) — why?~~
~~Economy tracks with carbon footprint~~
- ~~Two important periodic tables, leading into a discussion of CRMs, 3TG~~
- ~~Look for sustainability hotspots: metals and ceramics (two materials classes)~~
- Next meetings: will also consider plastics and composites

- Hotspots in steel making: mining, coal, ore to pig iron using coke
- Hotspots in ceramics: mining, drying, firing, except for glass – downcycling instead of recycling, lithium or other CRMs in advanced ceramics
- Approaches to decarbonize
- Rare earth metals – not rare in terms of abundance
- What's with Switzerland and gold?